

Tetracene Derivatives as Potential Red Emitters for Organic LEDs

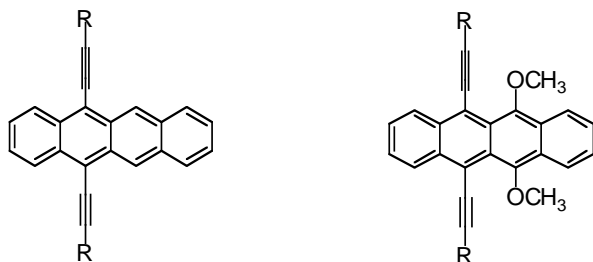
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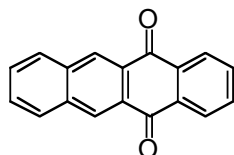
Supplementary Information

All materials and solvents were obtained from Aldrich Chemical Co. (Milwaukee, WI) or Acros Organics (Pittsburgh, PA) and were used without further purification unless otherwise noted. Commercially obtained alkynes were purchased from GFS Chemicals (Powell, OH). Moisture sensitive reactions were carried out in oven-dried glassware cooled under dry nitrogen. For solvents used in moisture sensitive reactions, diethylether was purified by passage through activated alumina, and dry THF was obtained from EM Science (DriSolv). All ^1H NMR spectra were obtained at 399.727 MHz, and all ^{13}C NMR spectra were recorded at 100.511 MHz using a Varian Inova 400 MHz NMR. ^1H NMR and ^{13}C NMR data are reported in parts per million (d) downfield from tetramethylsilane. Melting points were recorded using differential scanning calorimetry using a TA Instruments Differential Scanning Calorimeter model number DSC 2920. UV-Vis absorbance spectra were obtained using a Shimadzu UV-Vis recording spectrophotometer model UV-2501PC. Solution photoluminescence spectra and solid-state electroluminescence spectra were recorded using a Jobin Yvon-Spex Fluorolog-3 Fluorimeter model number FL3-11. Fluorescence Quantum Efficiencies were recorded using HPLC grade hexanes as the solvent, and the standard used was rubrene in hexanes, following a literature procedure.¹ Thin layer chromatography was performed on MERCK Silica Gel 60 thick layer plates. Silica gel chromatography was performed on Sorbent Technologies brand silica gel (32 – 63 mm, Standard Grade). Elemental analysis was performed by Complete Analysis Laboratories Inc., Parsippany, New Jersey. The evaporation system used to fabricate organic light-emitting diodes (OLEDs) was a modified system purchased from Cressington. Mass spectra for MALDI time-of-flight data were obtained using a Kratos Kompact SEQ spectrometer. Mass spectra for EI were obtained using a Finnigan Polaris Q using an ion trap mass analyzer.

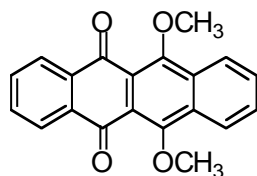
Synthetic Procedures and Characterization



General Procedure for the Synthesis of Tetracene Derivatives: One equivalent of the appropriate alkyne was dissolved in 20 mL of anhydrous diethyl ether in a 250 mL oven-dried round bottom flask under nitrogen atmosphere. 0.9 equivalents of *n*-butyl lithium (2.5 M solution in hexanes) were added by syringe and the solution was stirred at room temperature for 30 minutes. Then 0.33 equivalents of 5,12-tetracene quinone (**1**) or 5,12-dimethoxy-6,13-tetracene quinone (**2**) were added followed by 20 mL of anhydrous diethyl ether, and the reaction was allowed to stir at room temperature overnight. To the solution was then added 100 mL of saturated SnCl₂ in 10% aqueous HCl solution and THF was added to increased miscibility of the aqueous and organic layers. Upon addition, the flask was stirred under nitrogen for 30 minutes, and the reaction was monitored by TLC. Upon completion, the mixture was extracted with dichloromethane or hexanes and was washed with water. The solution was dried with MgSO₄ and was flushed through a cake of silica gel and was concentrated by rotary evaporation. The product was further purified by flash column chromatography with silica gel using either hexanes, 9:1 hexanes:dichloromethane, or 1:1 hexanes:dichloromethane, yielding a solid.

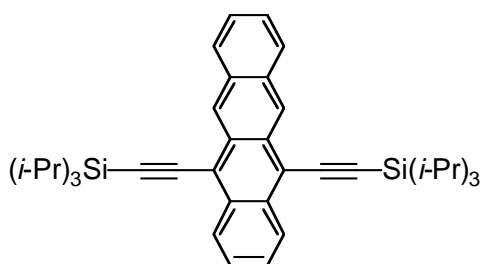


(1) 5,12-tetracene quinone. To an 500 mL round bottom flask under nitrogen atmosphere was added 1,4-naphthoquinone (1 g, 6.2 mmol), tetrabromo-*o*-xylene (10 g, 23.9 mmol), sodium iodide (23g, 163 mmol), calcium carbonate (5 g, 50 mmol), and *N,N*-dimethylformamide (75 mL). The mixture was sparged with nitrogen at room temperature for 30 minutes. The reaction was then brought to 55°C to stir for two days. The mixture was poured into 200 mL of water, and the yellow solid was filtered and washed with acetone. To remove calcium carbonate from the product, the solid was recrystallized with acetone. The product, a yellow solid, was obtained in 65% yield. The spectra data are consistent of a previous reported synthesis of this compound.²

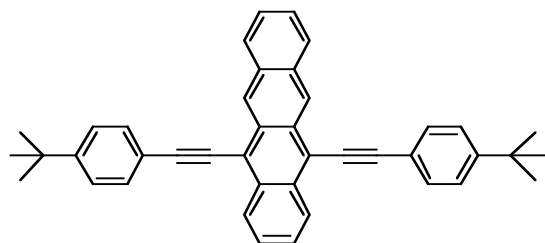


(2) 5,12-dimethoxy-6,13-tetracene quinone. 5,12-dihydroxy-6,13-tetracenequinone (1 g, 3.4 mmol), acetone (500 mL), and dioxane (150 mL) were combined in round bottom

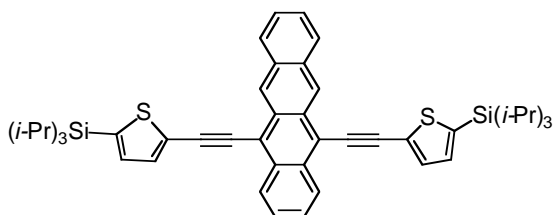
flask, and dimethyl sulfate (9.4 mL, 57 mmol) and anhydrous potassium carbonate (5 g, 39.4 mmol) were added. The solvents were sparged with nitrogen for 30 minutes and then heated to reflux under nitrogen for two days whereupon a yellow precipitate formed. The reaction mixture was poured into ice water to complete precipitation of the product, which was filtered and recrystallized from acetone. The product, a yellow solid, was isolated in 85% yield and had spectral data consistent with material reported in the literature.³



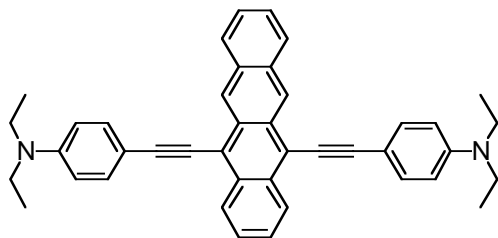
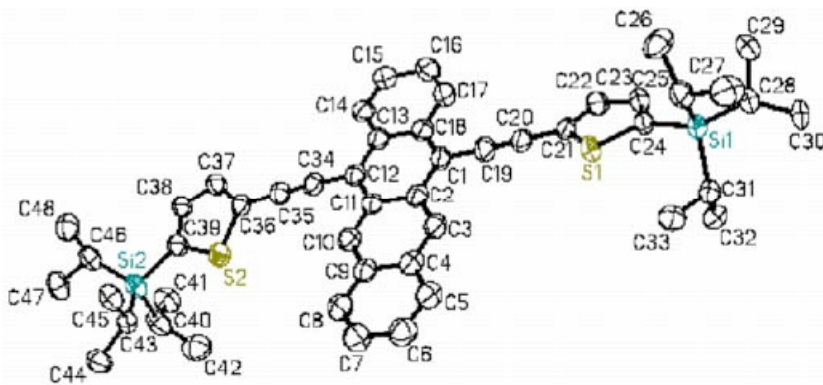
(3) 5,12-bis(triisopropylsilyl)ethynyl)tetracene was prepared following the general tetracene procedure using triisopropylsilyl acetylene and **1**. The product was recrystallized from acetone. ¹H NMR (CDCl₃) δ 1.35 (s, 42 H), 7.48 (dd, 2 H, *J* = 2.8, 6.4 Hz), 7.56 (dd, 2 H, *J* = 2.8, 6.8 Hz), 8.04 (dd, 2 H, *J* = 2.8, 6.8 Hz), 8.65 (dd, 2 H, *J* = 2.8, 6.4 Hz), 9.34 (s, 2 H). ¹³C NMR (CDCl₃) δ 11.81, 19.17, 104.15, 106.06, 118.81, 126.21, 126.52, 126.89, 127.62, 128.79, 130.56, 132.38, 132.83. MS (*m/z*, EI) 588 (*M*⁺, 100%), 545 (*M*⁺ - *i*-Pr). Anal Calcd for C₄₂H₃₆Si₂: C, 81.56; H, 8.89. Found C, 81.61; H, 9.15. Melting point: 130 °C. Fluorescence Quantum Efficiency: 0.79. Yield: 98%. UV-Vis (hexanes) λ_{nm} (log *e*): 291 (5.27), 316 (4.18), 332 (4.07), 351 (3.62), 466 (3.76), 498 (4.23), 535 (4.42).



(4) 5,12-bis(4-*t*-butylphenyl)ethynyl)tetracene was prepared using the general tetracene procedure using 4-*t*-butylphenylacetylene and **1**. The material was purified by chromatography on silica gel (hexanes eluent). ¹H NMR (CDCl₃) δ 1.42 (s, 18 H), 7.48 (dd, 2 H, *J* = 2.8, 6.4 Hz), 7.54 (d, 4 H, *J* = 6.8 Hz), 7.58 (dd, 2 H, *J* = 3.2, 6.8 Hz), 7.79 (d, 4H, *J* = 6.8 Hz), 8.12 (dd, 2 H, *J* = 3.2, 6.4 Hz), 8.70 (dd, 2 H, *J* = 3.2, 6.8 Hz), 9.32 (s, 2 H). ¹³C NMR (CDCl₃) δ 31.46, 35.18, 86.79, 103.64, 118.56, 120.78, 125.85, 126.13, 126.31, 126.71, 127.65, 128.80, 130.17, 131.72, 132.31, 132.45, 152.35. MS (*m/z*, EI) 540 (*M*⁺, 100%), 525 (*M*⁺ - CH₃). Anal Calcd for C₄₂H₃₆: C, 93.29; H, 6.71. Found C, 93.15; H, 6.62. Melting point: 132 °C. Fluorescence Quantum Efficiency: 0.81. Yield: 98%. UV-Vis (hexanes) λ_{nm} (log *e*): 290 (5.44), 357 (4.42), 485 (3.87), 514 (4.24), 556 (4.38).

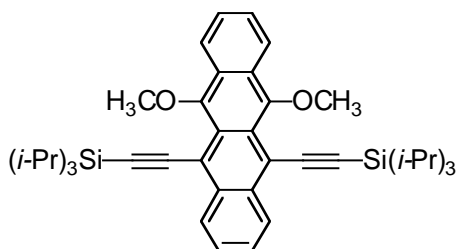


(5) 5,12-bis(5-triisopropylsilylthienylethynyl)tetracene was prepared using the general tetracene procedure using 5-triisopropylsilyl-2-ethynyl thiophene and **1**. ^1H NMR (CDCl_3) δ 1.19 (d, 18 H, $J = 7.2$ Hz), 1.44 (septet, 6 H, $J = 7.2$ Hz), 7.31 (d, 2 H, $J = 3.6$ Hz), 7.49 (dd, 2 H, $J = 2.8, 6.8$ Hz), 7.58 (dd, 2 H, $J = 3.2, 6.8$ Hz), 7.66 (d, 2 H, $J = 3.2$ Hz), 8.13 (dd, 2 H, $J = 3.2, 6.4$ Hz), 8.62 (dd, 2 H, $J = 3.2, 6.8$ Hz), 9.22 (s, 2 H). ^{13}C NMR (CDCl_3) δ 12.06, 18.81, 92.45, 96.84, 118.39, 126.24, 126.93, 127.60, 128.38, 128.83, 129.70, 129.87, 132.29, 132.39, 133.28, 135.96, 139.12. MS (m/z , EI) 752 (M^+), 710 (100%, $M^+ - i\text{-Pr}$). Anal Calcd for $\text{C}_{48}\text{H}_{56}\text{S}_2\text{Si}_2\text{O}_2$ (endoperoxide formed during shipment): C, 73.41; H, 7.18. Found C, 74.73; H, 7.28. Melting point: 187 °C. Fluorescence Quantum Efficiency: 0.69. Yield: 55%. UV-Vis (hexanes) λ_{nm} (log ϵ): 280 (5.33), 376 (4.36), 470 (3.62), 531 (4.28), 570 (4.41). Structure further confirmed by X-ray crystallographic analysis:

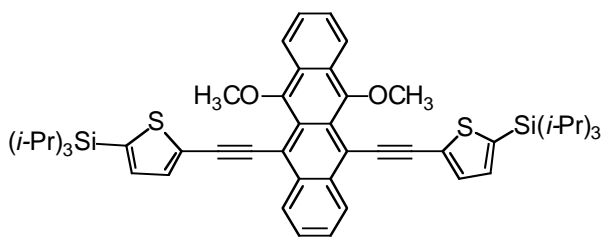


(6) 5,12-bis(4-(N,N-diethylamino)phenylethynyl)tetracene was prepared by the general procedure using 4-(diethylamino)phenylacetylene and **1**. The product was purified by recrystallization from 2-butanone. ^1H NMR (CDCl_3) δ 1.26 (t, 12 H, $J = 6.4$ Hz), 3.47 (q, 8 H, $J = 6.4$ Hz), 6.76 (d, 4 H, $J = 8.4$ Hz), 7.46 (dd, 2 H, $J = 3.2, 6.8$ Hz), 7.55 (dd, 2 H, $J = 3.2, 6.8$ Hz), 7.74 (d, 4 H, $J = 8.4$ Hz), 8.11 (dd, 2 H, $J = 3.2, 6.4$ Hz), 8.70 (dd, 2 H, $J = 3.2, 6.8$ Hz), 9.32 (s, 2H). ^{13}C NMR (CDCl_3) δ 12.81, 44.67, 85.71, 105.20, 109.50, 111.59, 118.50, 125.82, 126.29, 126.40, 127.83, 130.16, 132.13, 133.36, 148.04. MS (m/z , MALDI) = 570 (M^+). Anal Calcd for $\text{C}_{42}\text{H}_{36}\text{N}_2$: C, 88.69; H, 6.37; N, 4.92. Found

C, 87.99; H, 6.58; N, 5.15. Melting point: 250 °C (dec.). Fluorescence Quantum Efficiency: 0.10. Yield: 49%. UV-Vis (hexanes) λ_{nm} (log ϵ): 277 (5.96), 352 (4.37), 380 (4.42), 556 (4.51), 595 (4.61).



(7) 5,12-dimethoxy-6,11-bis(triisopropylsilyl)ethynyl)tetracene was prepared using the general tetracene procedure with triisopropylsilyl acetylene and **2**. The product was purified by recrystallization from acetone, yielding material suitable for X-ray analysis. ^1H NMR (CDCl_3) δ 1.32 (br. s, 42 H), 4.07 (s, 6 H), 7.48 (dd, 2 H, $J = 3.2, 6.8$), 7.55 (dd, 2 H, $J = 3.2, 6.8$ Hz), 8.33 (dd, 2 H, $J = 3.2, 6.8$ Hz), 8.90 (dd, 2 H, $J = 3.2, 6.8$ Hz). ^{13}C NMR (CDCl_3) δ 12.03, 19.12, 64.05, 106.02, 106.80, 115.81, 123.30, 124.95, 125.61, 126.08, 126.99, 127.92, 134.29, 149.94. MS (m/z , EI) 648 (100%, M^+), 633 (80%, $M^+ - \text{Me}$). Anal Calcd for $\text{C}_{42}\text{H}_{56}\text{O}_2\text{Si}_2$: C, 77.72; H, 8.69. Found C, 77.79; H, 9.01. Melting point: 124 °C. Fluorescence Quantum Efficiency: 0.19. Yield: 88%. UV-Vis (hexanes) λ_{nm} (log ϵ): 300 (5.29), 329 (4.12), 347 (3.95), 380 (3.62), 501 (3.81), 538 (4.19), 580 (4.32).



(8) 5,12-dimethoxy-6,11-bis(5-triisopropylsilylthienylethynyl)tetracene was prepared using the general tetracene procedure with 5-triisopropylsilyl-2-ethynyl thiophene and **2**. The product was purified by repeated chromatography on silica gel (hexanes:methylene chloride 9:1). ^1H NMR (CDCl_3) δ 1.19 (d, 36 H, $J = 7.2$ Hz), 1.43 (septet, 6 H, $J = 7.2$ Hz), 4.24 (s, 6H), 7.28 (d, 2 H, $J = 8$ Hz), 7.51 (dd, 2 H, $J = 3.2, 6.8$ Hz), 7.62 (d, 2 H $J = 3.2$ Hz), 7.64 (dd, 2 H, $J = 3.2, 6.8$ Hz), 8.40 (dd, 2 H, $J = 2.8, 6.4$ Hz), 8.84 (dd, 2 H, $J = 3.2, 6.8$ Hz). ^{13}C NMR (CDCl_3) δ 12.06, 18.81, 64.16, 96.44, 96.65, 115.22, 123.29, 124.70, 125.77, 126.27, 127.37, 127.52, 129.42, 132.80, 133.11, 136.01, 138.75, 149.89. MS (m/z , EI) 813 (50%, M^+), 786 (50%, $M^+ - 2 \text{ Me}$). Anal Calcd for $\text{C}_{50}\text{H}_{60}\text{S}_2\text{Si}_2\text{O}_4$ (endoperoxide formed during shipment): C, 71.04; H, 7.15. Found C, 71.51; H, 7.39. Melting point: 164 °C. Fluorescence Quantum Efficiency: 0.36. Yield: 17%. UV-Vis

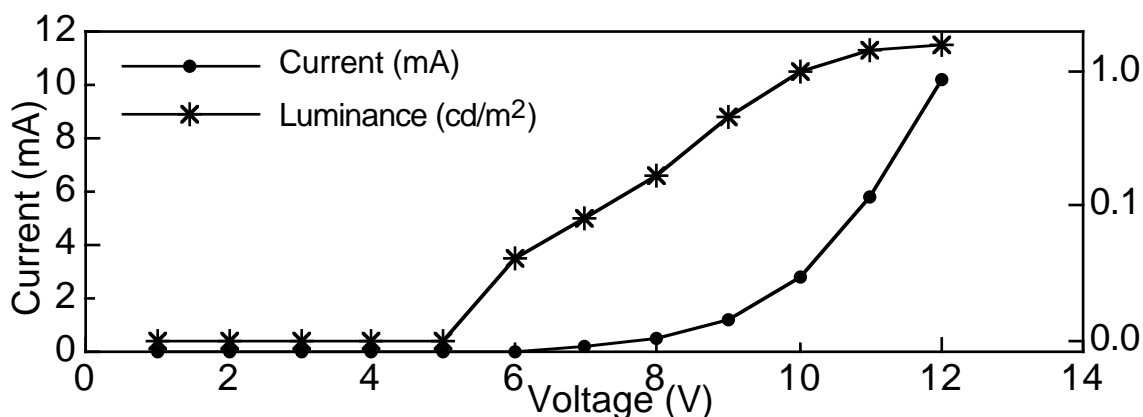
(hexanes) I_{nm} (log e): 303 (4.65), 389 (4.32), 412 (4.15), 532 (3.25), 571 (3.74), 617 (3.89).

Fabrication of OLEDs

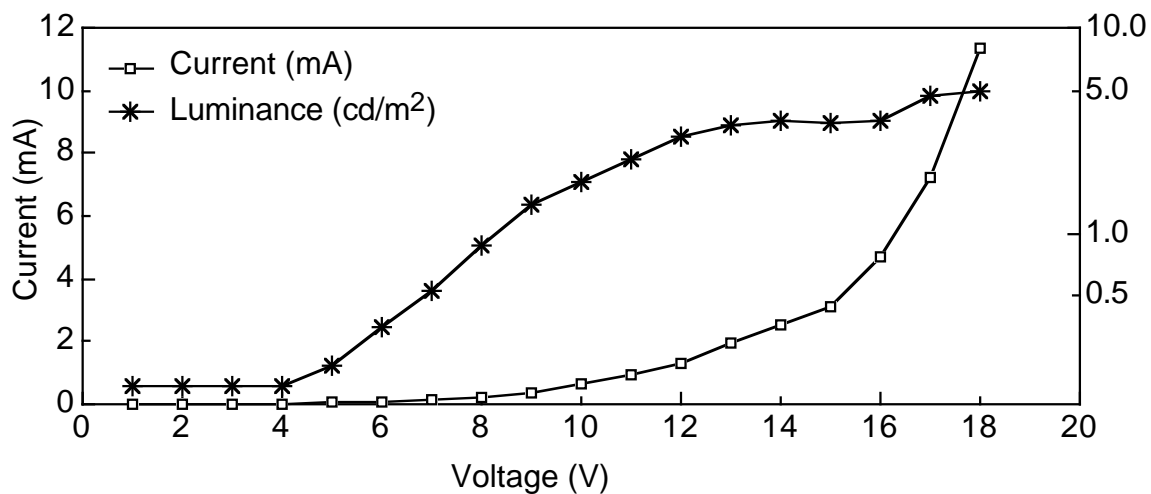
ITO glass was provided by Gentex Corporation. The ITO glass was patterned by using a shadow mask to spray lacquer onto the ITO glass, then removing the exposed ITO using an *aqua regia* bath (10:10:1::H₂O:HCl:HNO₃, 20 min.). The lacquer was removed with acetone. The remaining ITO was treated with sonication in a dilute *aqua regia* bath (2:7:25::HNO₃:HCl:H₂O) until the ITO turned from red-violet to blue-gold (approx. 20 min.). The ITO glass was treated with an RCA bath (1:1:5::H₂O₂:NH₄OH:H₂O, 30 min.). The substrates were washed with sonication in water (5 min.), isopropanol (5 min.), and acetone (5 min.) after which they were dried under nitrogen. The substrates were further dried in an oven (110°C, 15 min.).

Following the substrate preparation, layers of compounds were deposited via evaporation under a vacuum of 10⁻⁵ mbar. The layers used were the hole-transporting layer (TPD, 40 nm), the light-emitting layer (tetracene derivative, 30 nm), a buffer layer (LiF, 0.8 nm), and the cathode (aluminum, 60 – 100 nm). Current and luminance data were recorded for the OLEDs. Examples of the voltage-current-luminance data are shown below.

Current, Voltage, Luminance Data for OLED with Compound 3:



Current, Voltage, Luminance Data for OLED with Compound 7:



References:

1. Fery-Forgues, S.; Lavabre, D. *J. Chem. Ed.* **199**, 76, 1260.
2. Hanhela, P. J.; Paul, D. B. *Aust. J. Chem.* **1981**, 34, 1701.
3. Khanapure, S. P.; Reddy, R. T.; Biehl, E. R. *J. Org. Chem.* **1987**, 52, 5685.

Compound 3
¹H NMR

sao 5,12-bisIPsethynyltetracene

Pulse Sequence: s2pul

Solvent: CDCl₃

Ambient temperature

INNOVA-400 "Q10400"

PULSE SEQUENCE

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Pulse 1.2 degrees

Acq. time 0.042 sec

Width 6000.6 Hz

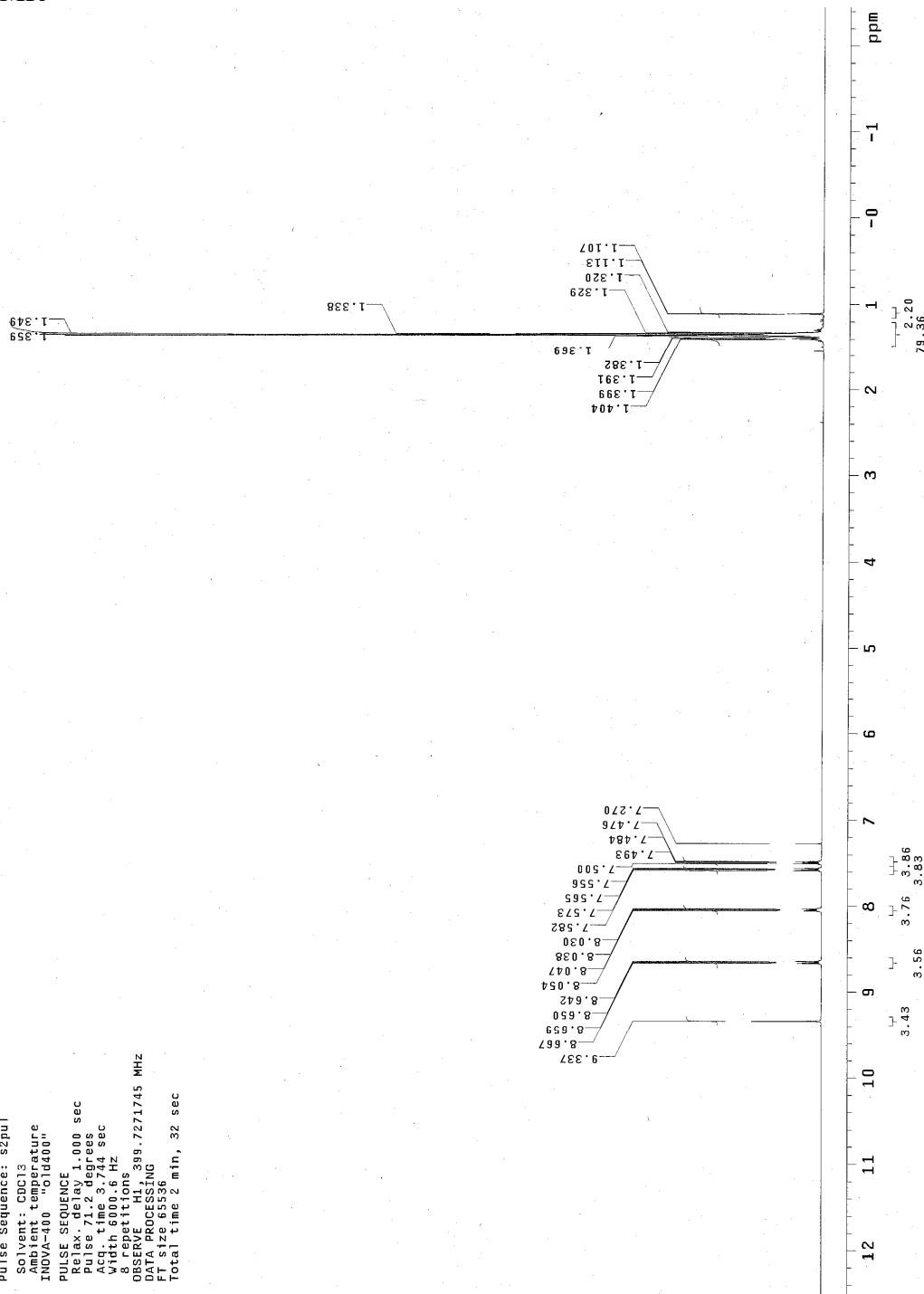
8 repetitions

OBSERVE H1, 399.7271745 MHz

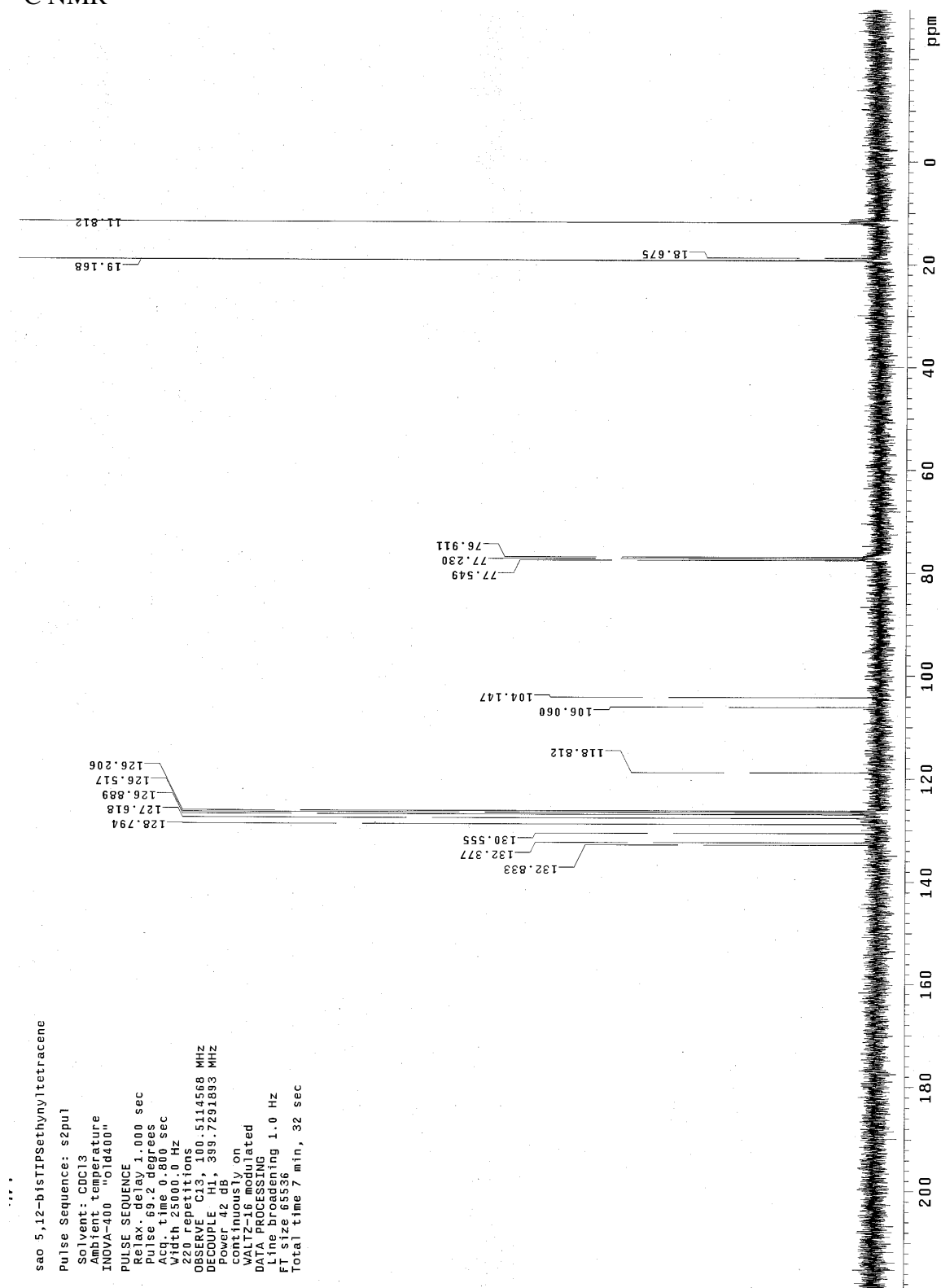
DATA PROCESSING

FT size 65536

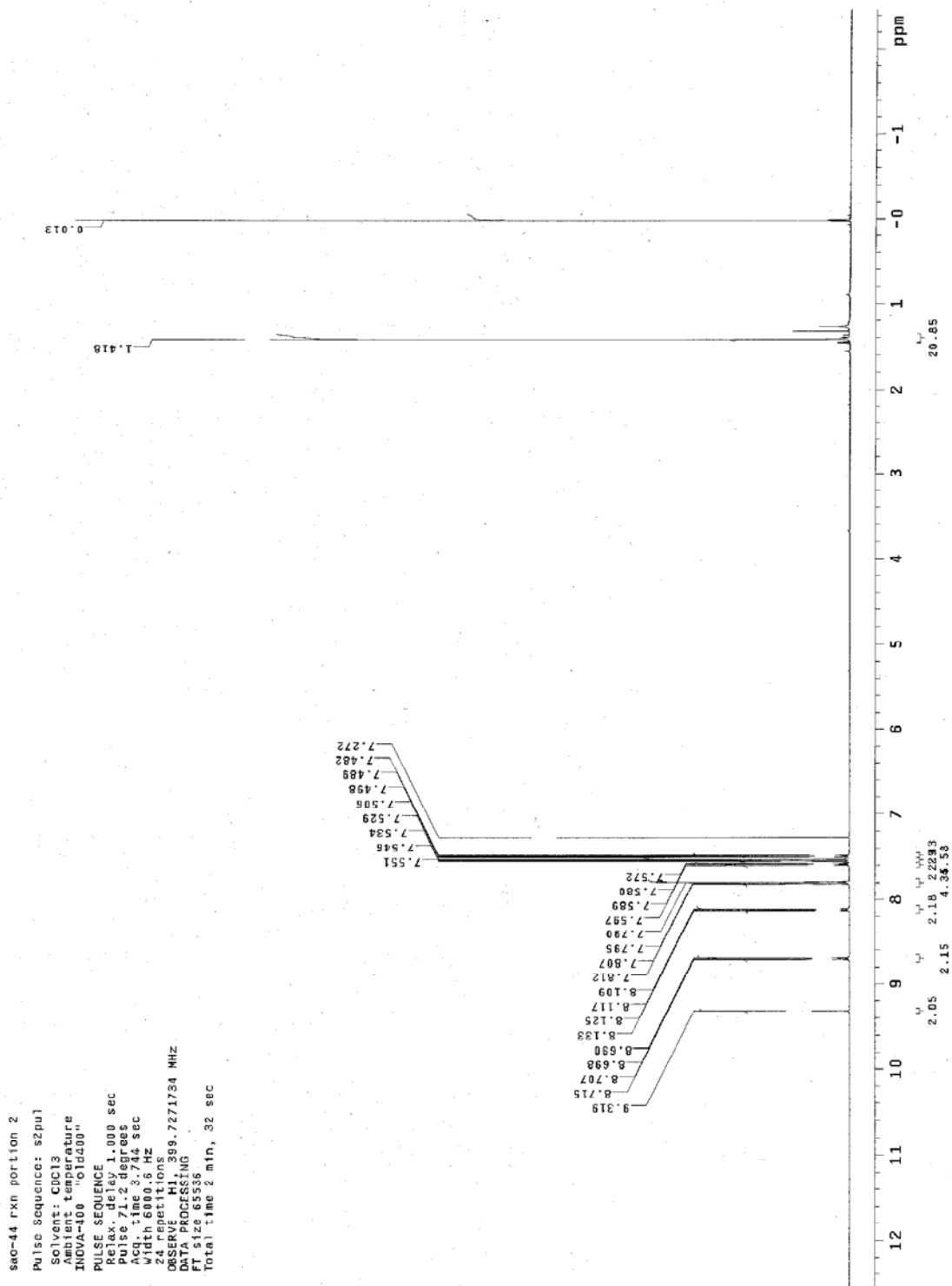
Total time 2 min, 32 sec



Compound 3
¹³C NMR



Compound 4
¹H NMR



Compound 4
¹³C NMR

seo bis-5,12-t-butylphenylethynyltetracene

Pulse Sequence: s2pu1

Solvent: CDCl₃

Ambient temperature

INDVA-400 "Q1d400"

PULSE SEQUENCE

Relax. delay 1.000 sec

Pulse 69.2 degrees

Acq. time 0.800 sec

Width 25000.0 Hz

220 repetitions

OBSERVE C13, 100.5114568 MHz

DECOUPLE H1, 399.7291893 MHz

Power 42 dB

continuously on

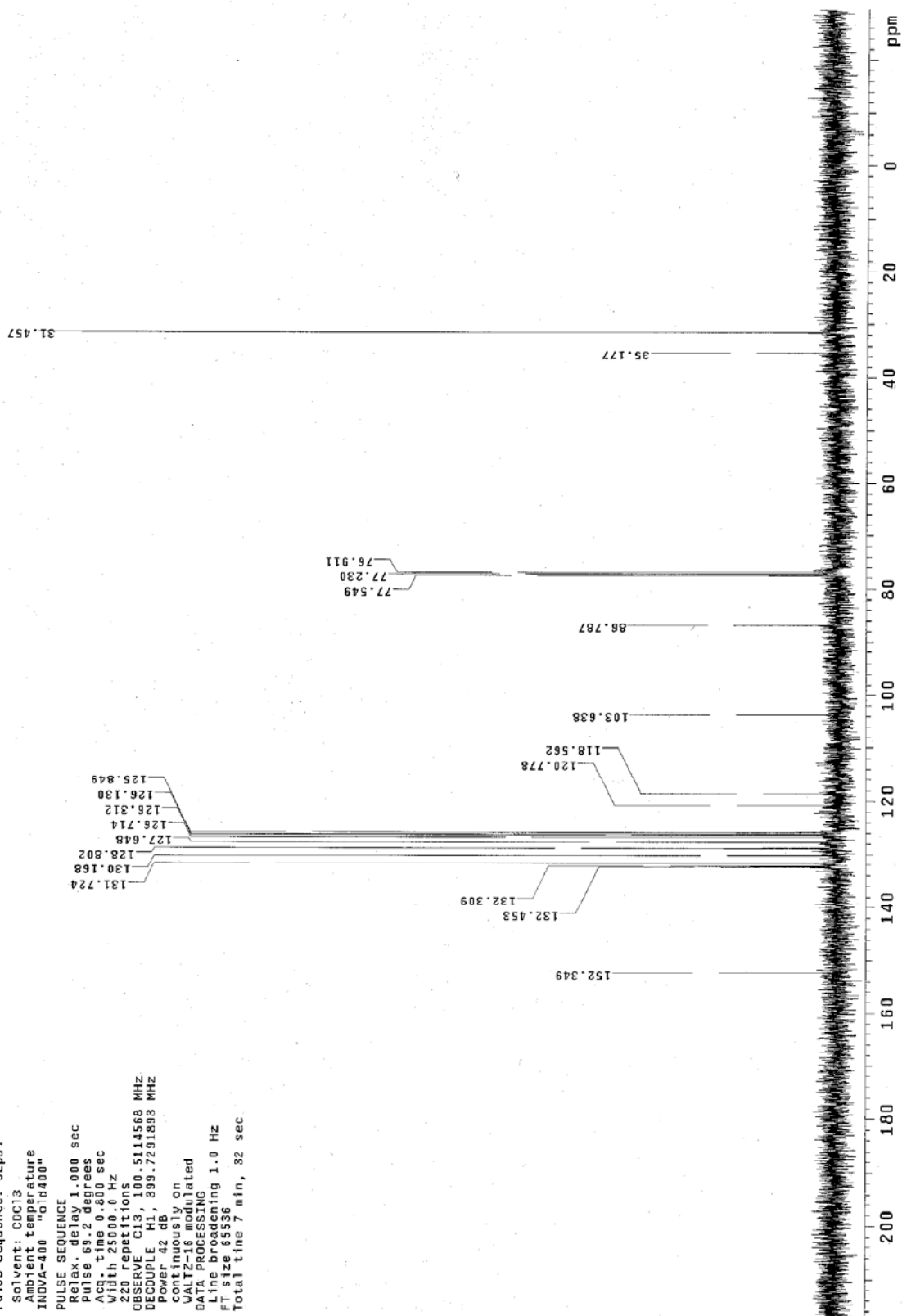
WALTZ-16 modulated

DATA PROCESSING

Line broadening 1.0 Hz

File size 85536

Total time 7 min, 32 sec



Compound 5
¹H NMR

sao 5,12-bis-(TIPSthienylethynyl)tetracene

Pulse Sequence: s2pul

Solvent: CCl₃

Ambient temperature

INNOVA-400 "Old400"

PULSE SEQUENCE

Relax. delay 1.000 sec

Pulse 71.2 degrees

Acq. time 3.744 sec

Width 6000.6 Hz

12 repetitions

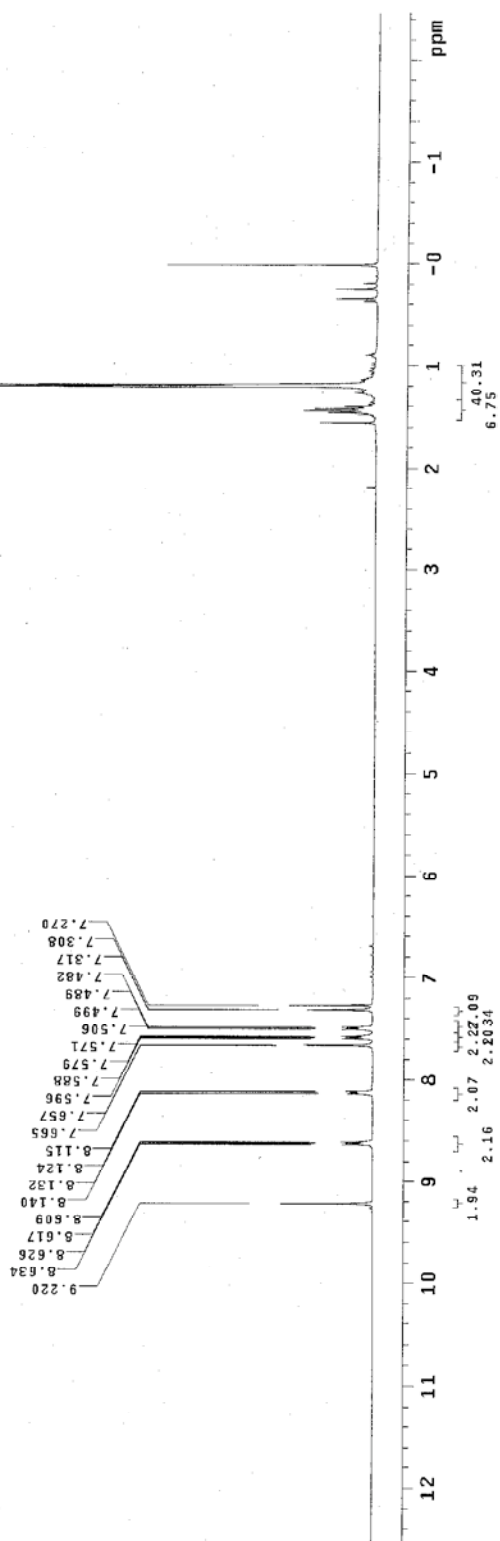
OBSERVE H1, 399.7271741 MHz

DATA PROCESSING

Time offsetting 1.0 Hz

FT size 65556

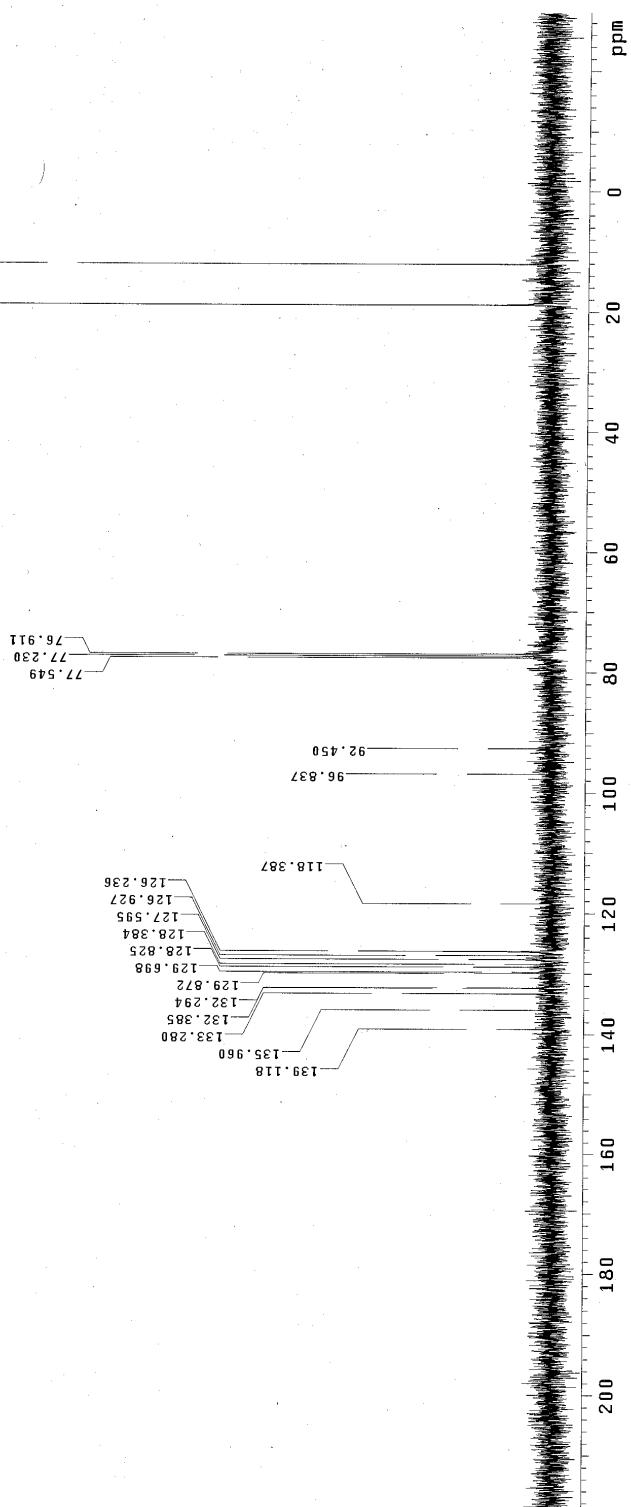
Total time 2 min, 32 sec



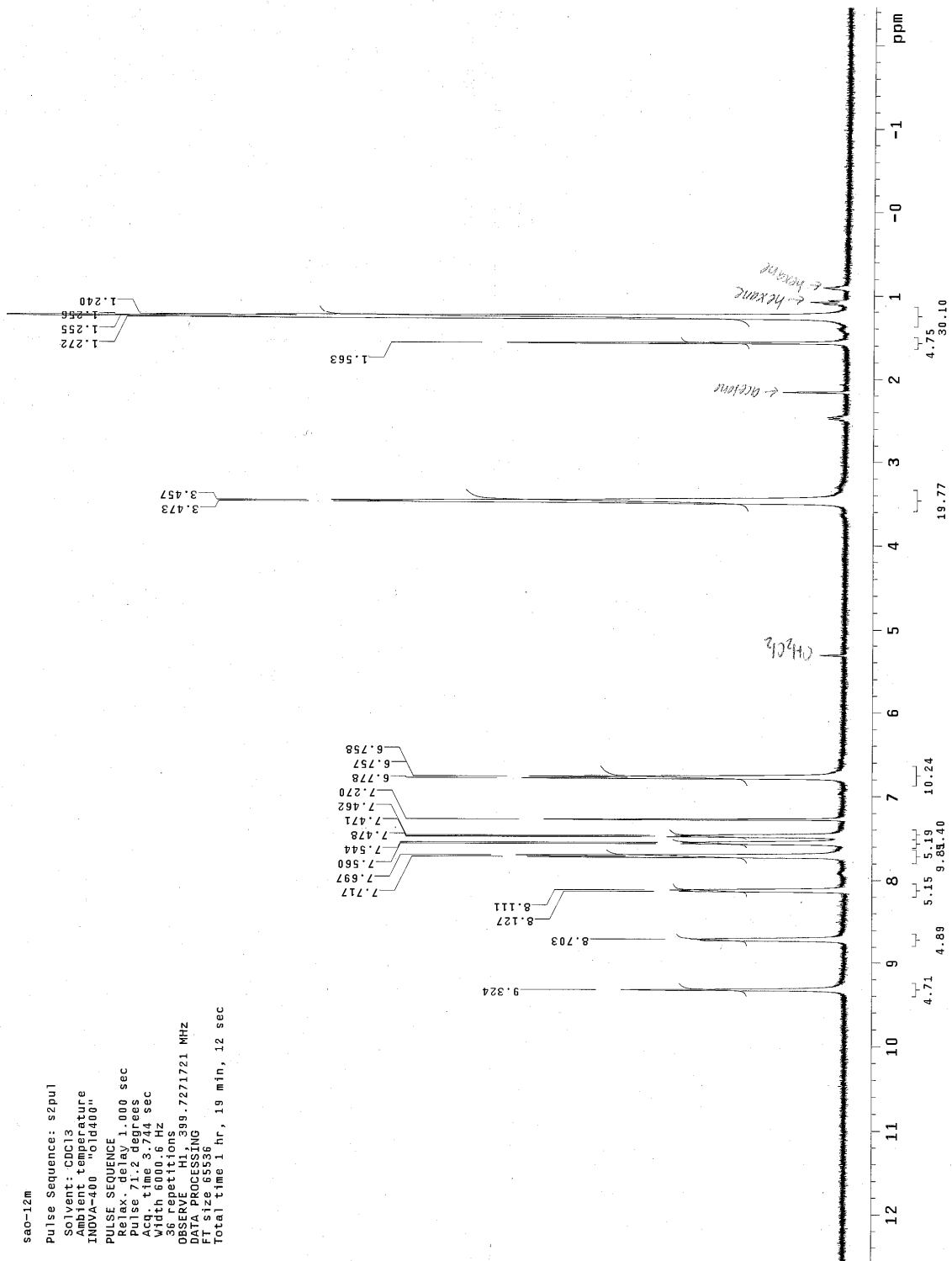
Compound 5
¹³C NMR

sao 5,12-bis-(TIPStienylethynyl)tetracene

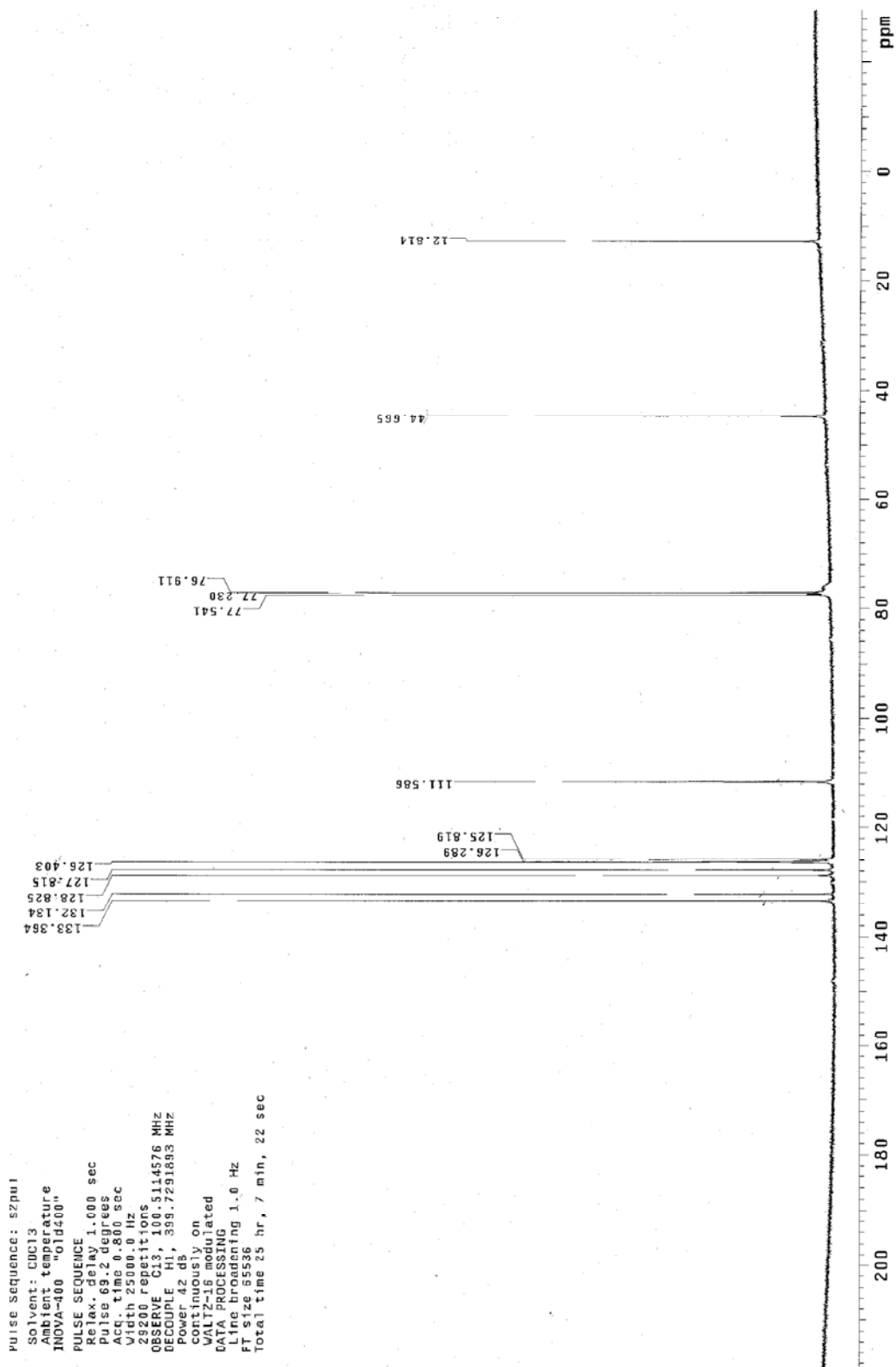
Pulse Sequence: s2pul
 Solvent: CDCl3
 Ambient temperature
 INOVA-400 "oid400"
 PULSE SEQUENCE
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 Pulse 69.2 degrees
 Acq. time 0.800 sec
 Width 25000.0 Hz
 OBSERVE 131.000 MHz
 DECOUPLE 1H, 399.7291893 MHz
 Power 42 dB
 continuously on
 VOLTAGE 16 modulated
 DATA PROCESSING
 Line broadening 1.0 Hz
 FT size 65536
 Total time 30 min, 8 sec.



Compound 6
¹H NMR



Compound 6
¹³C NMR



Compound 7
¹H NMR

sac 6,11-dimethoxy-5,12-bis-TripSethynyltetracene

Pulse Sequence: zgpg30

Solvent: CDCl₃

Ambient temperature

INOVA-400 "QNP400"

PULSE SEQUENCE

Relax. delay 1.000 sec

Pulse 71.2 degrees

Acq. time 3.744 sec

Width 6000.6 Hz

8 repetitions

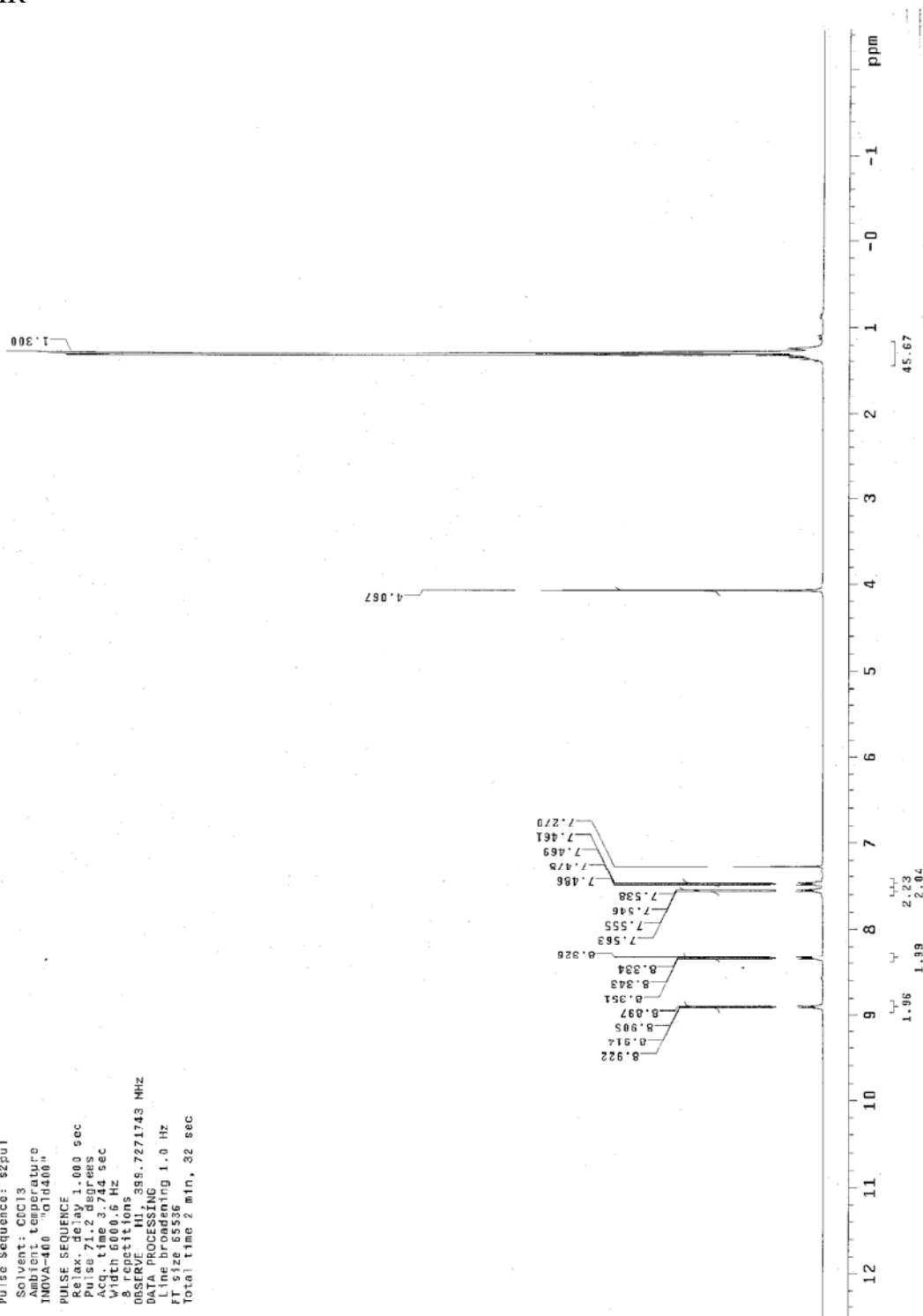
Observed FID

Processing

Line broadening 1.0 Hz

FT size 65536

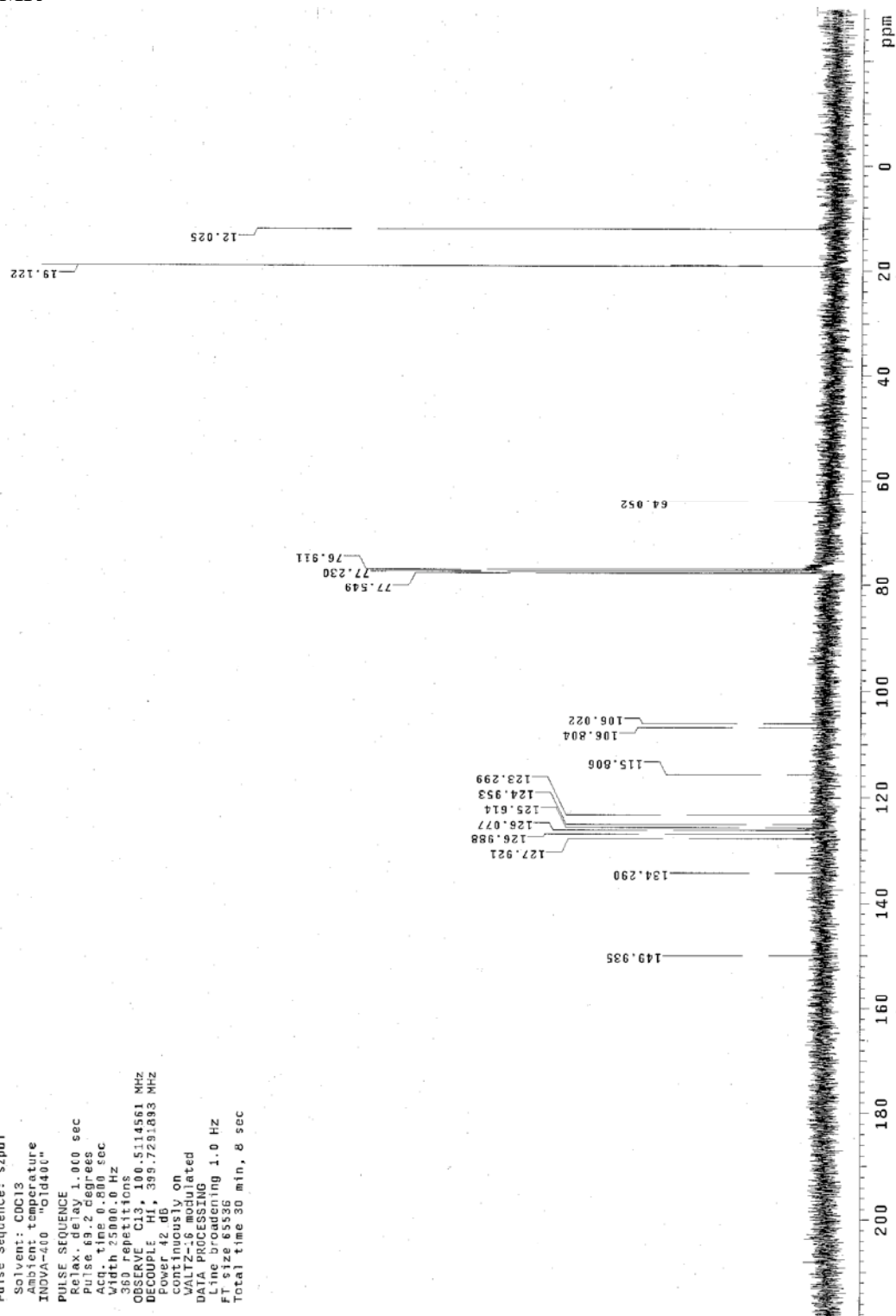
Total time 2 min, 32 sec



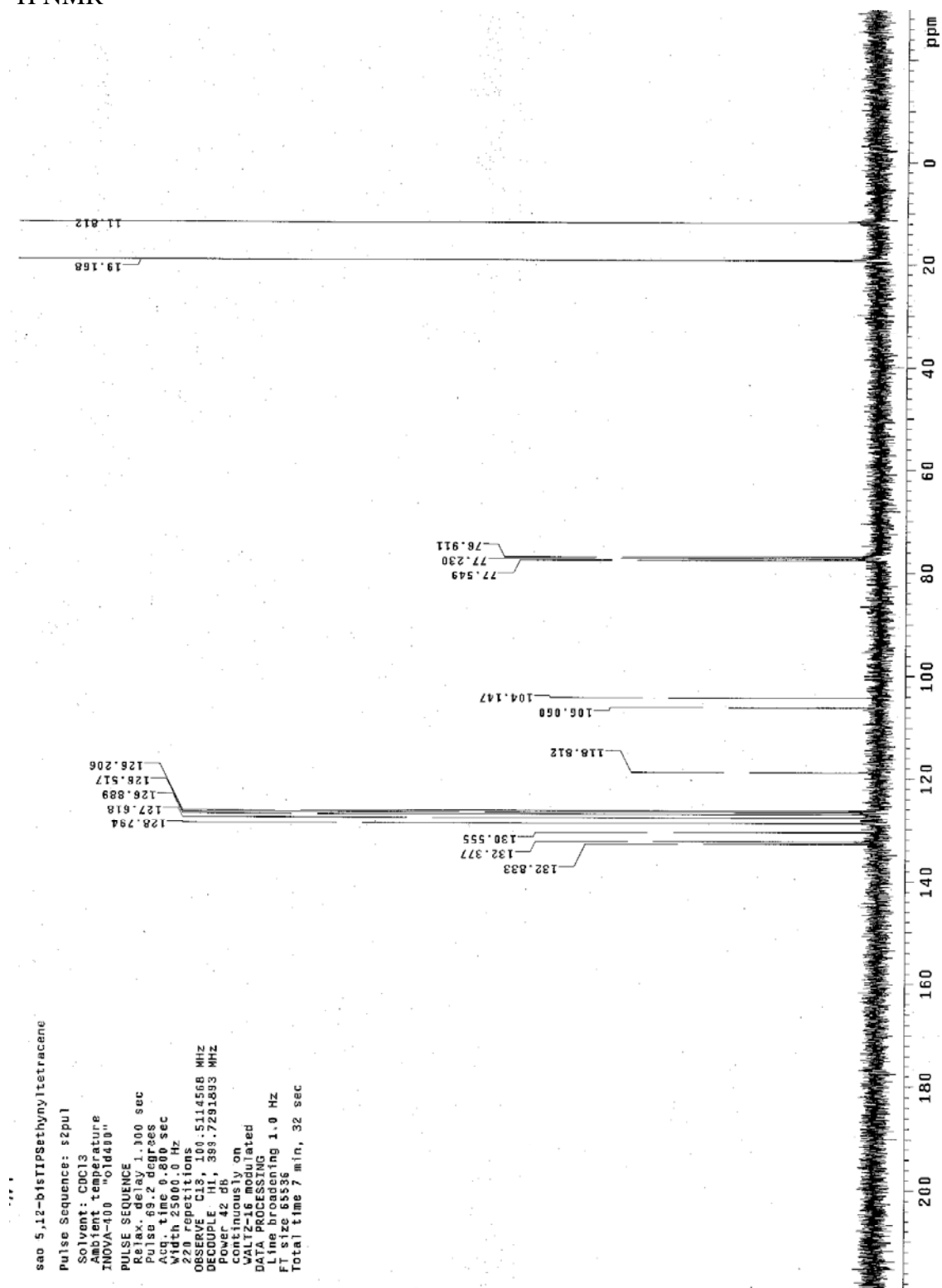
Compound 7
¹³C NMR

sao dimethoxy-bisTIPSethynyltetraquinone

Pulse Sequence: szpul
 Solvent: CDCl₃
 Ambient temperature
 INOVA-400 "G1d400"
 PULSE SEQUENCE
 Relax delay 1.000 sec
 Pulse 69.2 degrees
 Acq. time 0.800 sec
 Width 25000.0 Hz
 360 repetitions
 OBSERVE C13, 100.5114561 MHz
 DECOUPLE H1, 399.7291893 MHz
 Power 42 dB
 continuously on
 WALTZ-16 modulated
 DATA PROCESSING
 16 scans, averaging 1.0 Hz
 FT size 65536
 Total time 30 min, 8 sec



Compound 8
¹H NMR



Compound 8
¹³C NMR

