

# Tetracene Derivatives as Potential Red Emitters for Organic LEDs

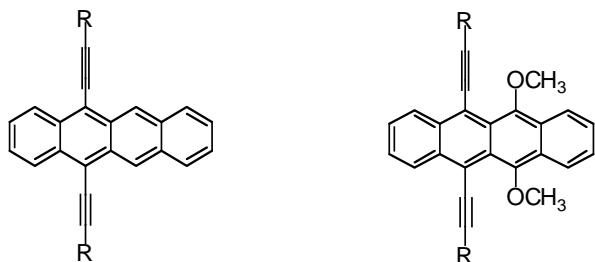
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Lexington, KY 40506-0055

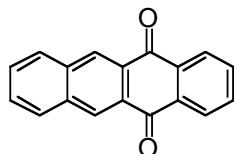
## Supplementary Information

All materials and solvents were obtained from Aldrich Chemical Co. (Milwaukee, WI) or Acros Organics (Pittsburgh, PA) and were used without further purification unless otherwise noted. Commercially obtained alkynes were purchased from GFS Chemicals (Powell, OH). Moisture sensitive reactions were carried out in oven-dried glassware cooled under dry nitrogen. For solvents used in moisture sensitive reactions, diethylether was purified by passage through activated alumina, and dry THF was obtained from EM Science (DriSolv). All <sup>1</sup>H NMR spectra were obtained at 399.727 MHz, and all <sup>13</sup>C NMR spectra were recorded at 100.511 MHz using a Varian Inova 400 MHz NMR. <sup>1</sup>H NMR and <sup>13</sup>C NMR data are reported in parts per million (d) downfield from tetramethylsilane. Melting points were recorded using differential scanning calorimetry using a TA Instruments Differential Scanning Calorimeter model number DSC 2920. UV-Vis absorbance spectra were obtained using a Shimadzu UV-Vis recording spectrophotometer model UV-2501PC. Solution photoluminescence spectra and solid-state electroluminescence spectra were recorded using a Jobin Yvon-Spex Fluorolog-3 Fluorimeter model number FL3-11. Fluorescence Quantum Efficiencies were recorded using HPLC grade hexanes as the solvent, and the standard used was rubrene in hexanes, following a literature procedure.<sup>1</sup> Thin layer chromatography was performed on MERCK Silica Gel 60 think layer plates. Silica gel chromatography was performed on Sorbent Technologies brand silica gel (32 – 63 mm, Standard Grade). Elemental analysis was performed by Complete Analysis Laboratories Inc., Parsippany, New Jersey. The evaporation system used to fabricate organic light-emitting diodes (OLEDs) was a modified system purchased from Cressington. Mass spectra for MALDI time-of-flight data were obtained using a Kratos Kompact SEQ spectrometer. Mass spectra for EI were obtained using a Finnigan Polaris Q using an ion trap mass analyzer.

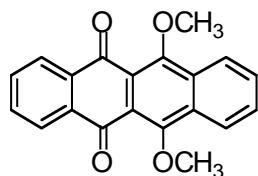
## Synthetic Procedures and Characterization



**General Procedure for the Synthesis of Tetracene Derivatives:** One equivalent of the appropriate alkyne was dissolved in 20 mL of anhydrous diethyl ether in a 250 mL oven-dried round bottom flask under nitrogen atmosphere. 0.9 equivalents of *n*-butyl lithium (2.5 M solution in hexanes) were added by syringe and the solution was stirred at room temperature for 30 minutes. Then 0.33 equivalents of 5,12-tetracene quinone (**1**) or 5,12-dimethoxy-6,13-tetracene quinone (**2**) were added followed by 20 mL of anhydrous diethyl ether, and the reaction was allowed to stir at room temperature overnight. To the solution was then added 100 mL of saturated  $\text{SnCl}_2$  in 10% aqueous HCl solution and THF was added to increased miscibility of the aqueous and organic layers. Upon addition, the flask was stirred under nitrogen for 30 minutes, and the reaction was monitored by TLC. Upon completion, the mixture was extracted with dichloromethane or hexanes and was washed with water. The solution was dried with  $\text{MgSO}_4$  and was flushed through a cake of silica gel and was concentrated by rotary evaporation. The product was further purified by flash column chromatography with silica gel using either hexanes, 9:1 hexanes:dichloromethane, or 1:1 hexanes:dichloromethane, yielding a solid.

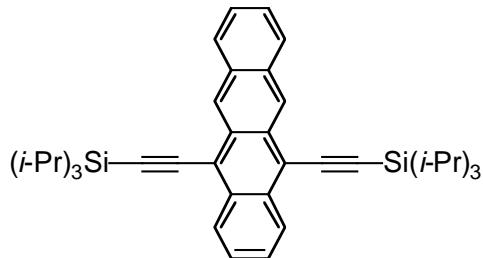


**(1) 5,12-tetracene quinone.** To an 500 mL round bottom flask under nitrogen atmosphere was added 1,4-napthoquinone (1 g, 6.2 mmol), tetrabromo-*o*-xylene (10 g, 23.9 mmol), sodium iodide (23g, 163 mmol), calcium carbonate (5 g, 50 mmol), and *N,N*-dimethylformamide (75 mL). The mixture was sparged with nitrogen at room temperature for 30 minutes. The reaction was then brought to 55°C to stir for two days. The mixture was poured into 200 mL of water, and the yellow solid was filtered and washed with acetone. To remove calcium carbonate from the product, the solid was recrystallized with acetone. The product, a yellow solid, was obtained in 65% yield. The spectra data are consistent of a previous reported synthesis of this compound.<sup>2</sup>

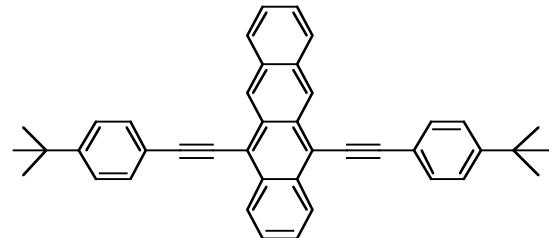


**(2) 5,12-dimethoxy-6,13-tetracene quinone.** 5,12-dihydroxy-6,13-tetracenequinone (1 g, 3.4 mmol), acetone (500 mL), and dioxane (150 mL) were combined in round bottom

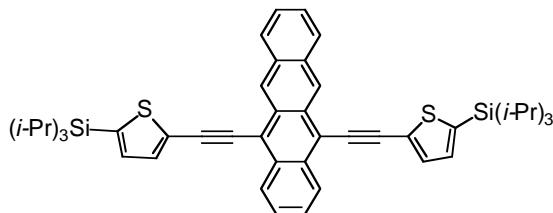
flask, and dimethyl sulfate (9.4 mL, 57 mmol) and anhydrous potassium carbonate (5 g, 39.4 mmol) were added. The solvents were sparged with nitrogen for 30 minutes and then heated to reflux under nitrogen for two days whereupon a yellow precipitate formed. The reaction mixture was poured into ice water to complete precipitation of the product, which was filtered and recrystallized from acetone. The product, a yellow solid, was isolated in 85% yield and had spectral data consistent with material reported in the literature.<sup>3</sup>



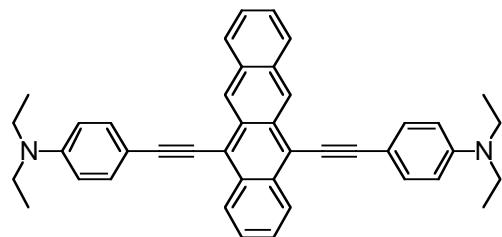
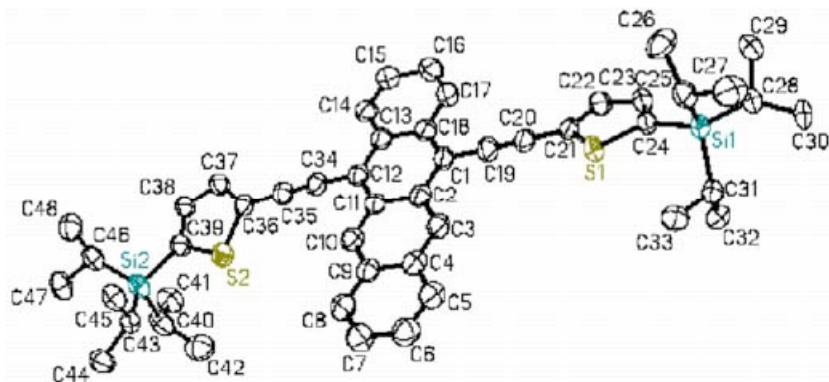
**(3)** 5,12-bis(triisopropylsilyl)ethynyl)tetracene was prepared following the general tetracene procedure using triisopropylsilyl acetylene and **1**. The product was recrystallized from acetone. <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.35 (s, 42 H), 7.48 (dd, 2 H, *J* = 2.8, 6.4 Hz), 7.56 (dd, 2 H, *J* = 2.8, 6.8 Hz), 8.04 (dd, 2 H, *J* = 2.8, 6.8 Hz), 8.65 (dd, 2 H, *J* = 2.8, 6.4 Hz), 9.34 (s, 2 H). <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 11.81, 19.17, 104.15, 106.06, 118.81, 126.21, 126.52, 126.89, 127.62, 128.79, 130.56, 132.38, 132.83. MS (m/z, EI) 588 (M+, 100%), 545 (M+ - *i*-Pr). Anal Calcd for C<sub>42</sub>H<sub>36</sub>Si<sub>2</sub>: C, 81.56; H, 8.89. Found C, 81.61; H, 9.15. Melting point: 130 °C. Fluorescence Quantum Efficiency: 0.79. Yield: 98%. UV-Vis (hexanes)  $\lambda_{\text{nm}}$  (log *e*): 291 (5.27), 316 (4.18), 332 (4.07), 351 (3.62), 466 (3.76), 498 (4.23), 535 (4.42).



**(4)** 5,12-bis(4-*t*-butylphenylethynyl)tetracene was prepared using the general tetracene procedure using 4-*t*-butylphenylacetylene and **1**. The material was purified by chromatography on silica gel (hexanes eluent). <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.42 (s, 18 H), 7.48 (dd, 2 H, *J* = 2.8, 6.4 Hz), 7.54 (d, 4 H, *J* = 6.8 Hz), 7.58 (dd, 2 H, *J* = 3.2, 6.8 Hz), 7.79 (d, 4H, *J* = 6.8 Hz), 8.12 (dd, 2 H, *J* = 3.2, 6.4 Hz), 8.70 (dd, 2 H, *J* = 3.2, 6.8 Hz), 9.32 (s, 2 H). <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 31.46, 35.18, 86.79, 103.64, 118.56, 120.78, 125.85, 126.13, 126.31, 126.71, 127.65, 128.80, 130.17, 131.72, 132.31, 132.45, 152.35. MS (m/z, EI) 540 (M+, 100%), 525 (M+ - CH<sub>3</sub>). Anal Calcd for C<sub>42</sub>H<sub>36</sub>: C, 93.29; H, 6.71. Found C, 93.15; H, 6.62. Melting point: 132 °C. Fluorescence Quantum Efficiency: 0.81. Yield: 98%. UV-Vis (hexanes)  $\lambda_{\text{nm}}$  (log *e*): 290 (5.44), 357 (4.42), 485 (3.87), 514 (4.24), 556 (4.38).

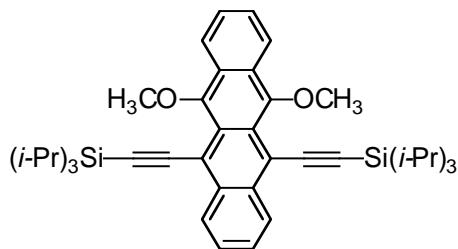


**(5)** 5,12-bis(5-triisopropylsilylthienylethynyl)tetracene was prepared using the general tetracene procedure using 5-triisopropylsilyl-2-ethynyl thiophene and **1**.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  1.19 (d, 18 H,  $J$  = 7.2 Hz), 1.44 (septet, 6 H,  $J$  = 7.2 Hz), 7.31 (d, 2 H,  $J$  = 3.6 Hz), 7.49 (dd, 2 H,  $J$  = 2.8, 6.8 Hz), 7.58 (dd, 2 H,  $J$  = 3.2, 6.8 Hz), 7.66 (d, 2 H,  $J$  = 3.2 Hz), 8.13 (dd, 2 H,  $J$  = 3.2, 6.4 Hz), 8.62 (dd, 2 H,  $J$  = 3.2, 6.8 Hz), 9.22 (s, 2 H).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  12.06, 18.81, 92.45, 96.84, 118.39, 126.24, 126.93, 127.60, 128.38, 128.83, 129.70, 129.87, 132.29, 132.39, 133.28, 135.96, 139.12. MS (m/z, EI) 752 (M+), 710 (100%, M+ - *i*-Pr). Anal Calcd for  $\text{C}_{48}\text{H}_{56}\text{S}_2\text{Si}_2\text{O}_2$  (endoperoxide formed during shipment): C, 73.41; H, 7.18. Found C, 74.73; H, 7.28. Melting point: 187 °C. Fluorescence Quantum Efficiency: 0.69. Yield: 55%. UV-Vis (hexanes)  $\lambda_{\text{nm}}$  (log  $\epsilon$ ): 280 (5.33), 376 (4.36), 470 (3.62), 531 (4.28), 570 (4.41). Structure further confirmed by X-ray crystallographic analysis:

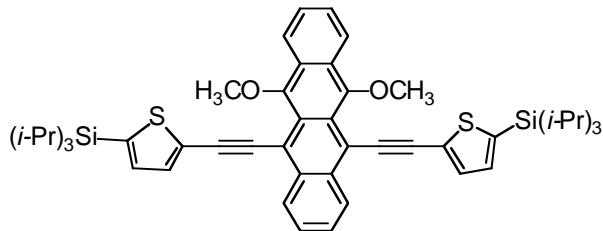


**(6)** 5,12-bis(4-(N,N-diethylamino)phenylethynyl)tetracene was prepared by the general procedure using 4-(diethylamino)phenylacetylene and **1**. The product was purified by recrystallization from 2-butanone.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  1.26 (t, 12 H,  $J$  = 6.4 Hz), 3.47 (q, 8 H,  $J$  = 6.4 Hz), 6.76 (d, 4 H,  $J$  = 8.4 Hz), 7.46 (dd, 2 H,  $J$  = 3.2, 6.8 Hz), 7.55 (dd, 2 H,  $J$  = 3.2, 6.8 Hz), 7.74 (d, 4 H,  $J$  = 8.4 Hz), 8.11 (dd, 2 H,  $J$  = 3.2, 6.4 Hz), 8.70 (dd, 2 H,  $J$  = 3.2, 6.8 Hz), 9.32 (s, 2H).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  12.81, 44.67, 85.71, 105.20, 109.50, 111.59, 118.50, 125.82, 126.29, 126.40, 127.83, 130.16, 132.13, 133.36, 148.04. MS (m/z, MALDI) = 570 (M+). Anal Calcd for  $\text{C}_{42}\text{H}_{36}\text{N}_2$ : C, 88.69; H, 6.37; N, 4.92. Found

C, 87.99; H, 6.58; N, 5.15. Melting point: 250 °C (dec.). Fluorescence Quantum Efficiency: 0.10. Yield: 49%. UV-Vis (hexanes)  $\lambda_{\text{nm}}$  (log  $\epsilon$ ): 277 (5.96), 352 (4.37), 380 (4.42), 556 (4.51), 595 (4.61).



(7) 5,12-dimethoxy-6,11-bis(triisopropylsilyl)ethynyl)tetracene was prepared using the general tetracene procedure with triisopropylsilyl acetylene and **2**. The product was purified by recrystallization from acetone, yielding material suitable for X-ray analysis.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  1.32 (br. s, 42 H), 4.07 (s, 6 H), 7.48 (dd, 2 H,  $J$  = 3.2, 6.8), 7.55 (dd, 2 H,  $J$  = 3.2, 6.8 Hz), 8.33 (dd, 2 H,  $J$  = 3.2, 6.8 Hz), 8.90 (dd, 2 H,  $J$  = 3.2, 6.8 Hz).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  12.03, 19.12, 64.05, 106.02, 106.80, 115.81, 123.30, 124.95, 125.61, 126.08, 126.99, 127.92, 134.29, 149.94. MS (m/z, EI) 648 (100%, M+), 633 (80%, M+ - Me). Anal Calcd for  $\text{C}_{42}\text{H}_{56}\text{O}_2\text{Si}_2$ : C, 77.72; H, 8.69. Found C, 77.79; H, 9.01. Melting point: 124 °C. Fluorescence Quantum Efficiency: 0.19. Yield: 88%. UV-Vis (hexanes)  $\lambda_{\text{nm}}$  (log  $\epsilon$ ): 300 (5.29), 329 (4.12), 347 (3.95), 380 (3.62), 501 (3.81), 538 (4.19), 580 (4.32).



(8) 5,12-dimethoxy-6,11-bis(5-triisopropylsilylthienyl)ethynyl)tetracene was prepared using the general tetracene procedure with 5-triisopropylsilyl-2-ethynyl thiophene and **2**. The product was purified by repeated chromatography on silica gel (hexanes:methylene chloride 9:1).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  1.19 (d, 36 H,  $J$  = 7.2 Hz), 1.43 (septet, 6 H,  $J$  = 7.2 Hz), 4.24 (s, 6H), 7.28 (d, 2 H,  $J$  = 8 Hz), 7.51 (dd, 2 H,  $J$  = 3.2, 6.8 Hz), 7.62 (d, 2 H  $J$  = 3.2 Hz), 7.64 (dd, 2 H,  $J$  = 3.2, 6.8 Hz), 8.40 (dd, 2 H,  $J$  = 2.8, 6.4 Hz), 8.84 (dd, 2 H,  $J$  = 3.2, 6.8 Hz).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  12.06, 18.81, 64.16, 96.44, 96.65, 115.22, 123.29, 124.70, 125.77, 126.27, 127.37, 127.52, 129.42, 132.80, 133.11, 136.01, 138.75, 149.89. MS (m/z, EI) 813 (50%, M+), 786 (50%, M+ - 2 Me). Anal Calcd for  $\text{C}_{50}\text{H}_{60}\text{S}_2\text{Si}_2\text{O}_4$  (endoperoxide formed during shipment): C, 71.04; H, 7.15. Found C, 71.51; H, 7.39. Melting point: 164 °C. Fluorescence Quantum Efficiency: 0.36. Yield: 17%. UV-Vis

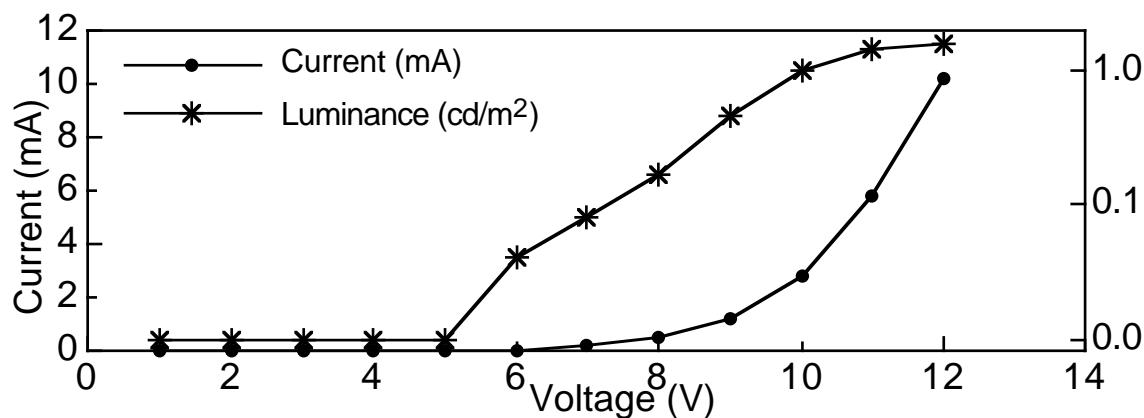
(hexanes)  $I_{nm}$  (log e): 303 (4.65), 389 (4.32), 412 (4.15), 532 (3.25), 571 (3.74), 617 (3.89).

### Fabrication of OLEDs

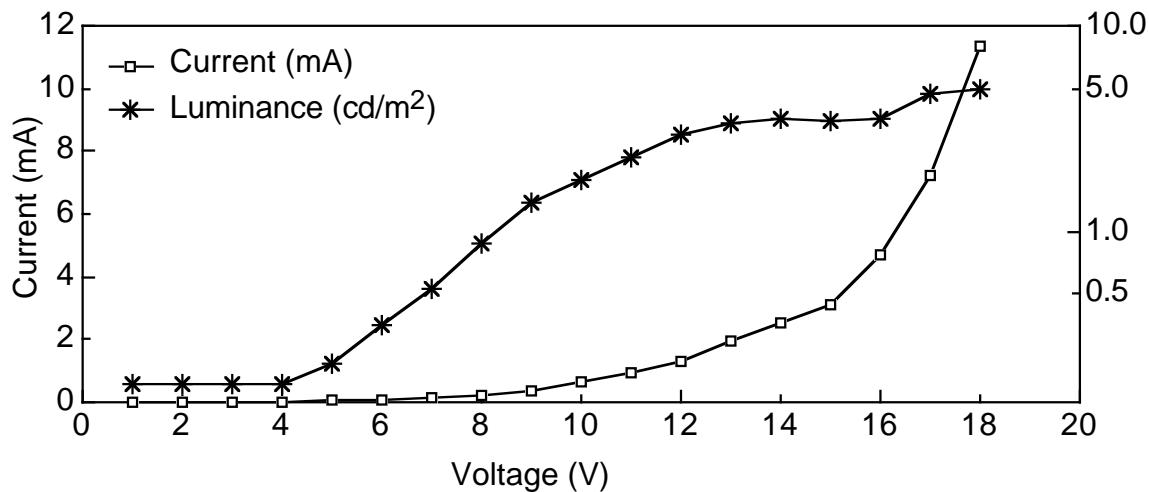
ITO glass was provided by Gentex Corporation. The ITO glass was patterned by using a shadow mask to spray lacquer onto the ITO glass, then removing the exposed ITO using an *aqua regia* bath (10:10:1::H<sub>2</sub>O:HCl:HNO<sub>3</sub>, 20 min.). The lacquer was removed with acetone. The remaining ITO was treated with sonication in a dilute *aqua regia* bath (2:7:25::HNO<sub>3</sub>:HCl:H<sub>2</sub>O) until the ITO turned from red-violet to blue-gold (approx. 20 min.). The ITO glass was treated with an RCA bath (1:1:5::H<sub>2</sub>O<sub>2</sub>:NH<sub>4</sub>OH:H<sub>2</sub>O, 30 min.). The substrates were washed with sonication in water (5 min.), isopropanol (5 min.), and acetone (5 min.) after which they were dried under nitrogen. The substrates were further dried in an oven (110°C, 15 min.).

Following the substrate preparation, layers of compounds were deposited via evaporation under a vacuum of 10<sup>-5</sup> mbar. The layers used were the hole-transporting layer (TPD, 40 nm), the light-emitting layer (tetracene derivative, 30 nm), a buffer layer (LiF, 0.8 nm), and the cathode (aluminum, 60 – 100 nm). Current and luminance data were recorded for the OLEDs. Examples of the voltage-current-luminance data are shown below.

#### Current, Voltage, Luminance Data for OLED with Compound 3:



#### Current, Voltage, Luminance Data for OLED with Compound 7:



References:

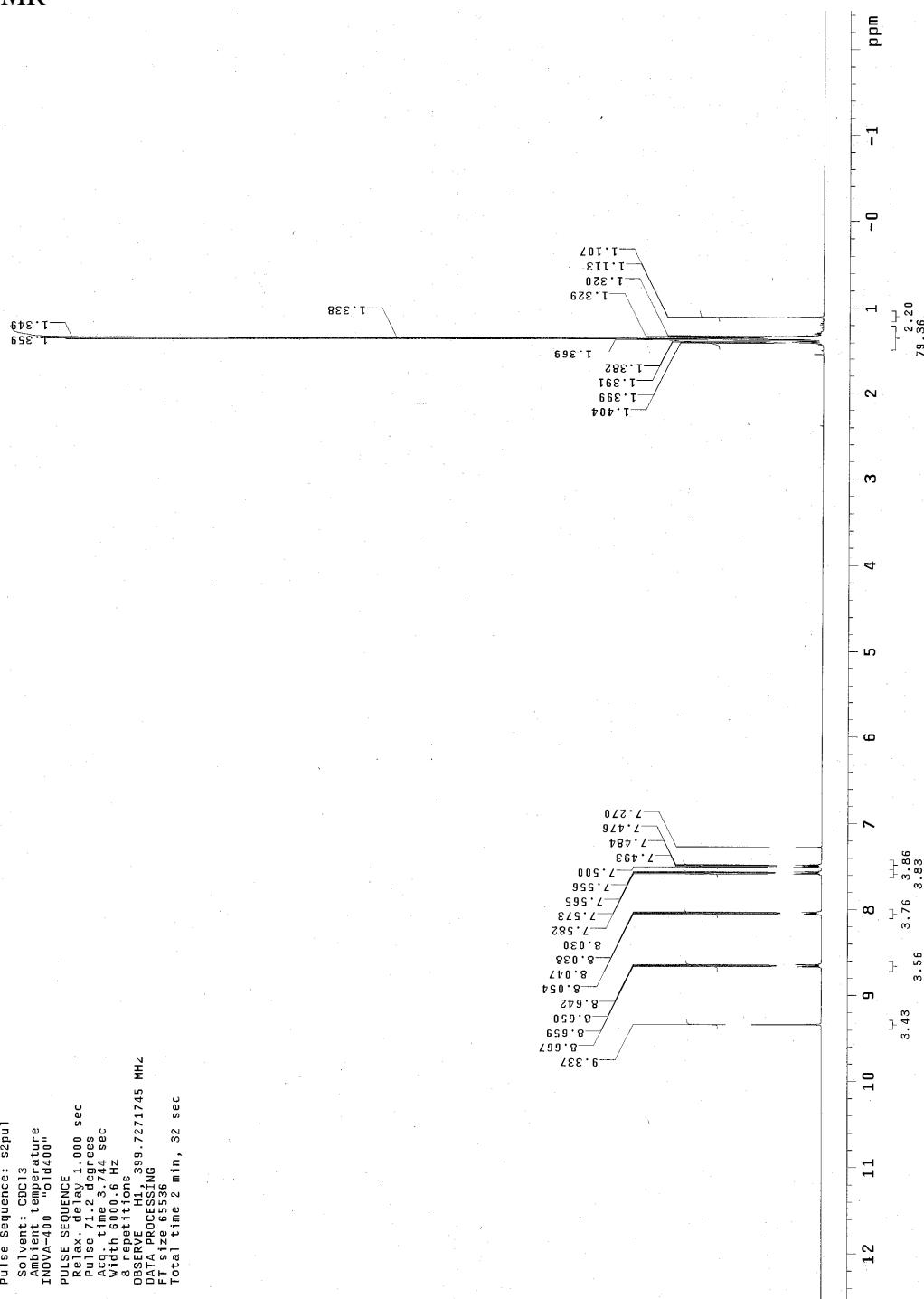
1. Fery-Forgues, S.; Lavabre, D. *J. Chem. Ed.* **199**, 76, 1260.
2. Hanhela, P. J.; Paul, D. B. *Aust. J. Chem.* **1981**, 34, 1701.
3. Khanapure, S. P.; Reddy, R. T.; Biehl, E. R. *J. Org. Chem.* **1987**, 52, 5685.

## Compound 3 <sup>1</sup>H NMR

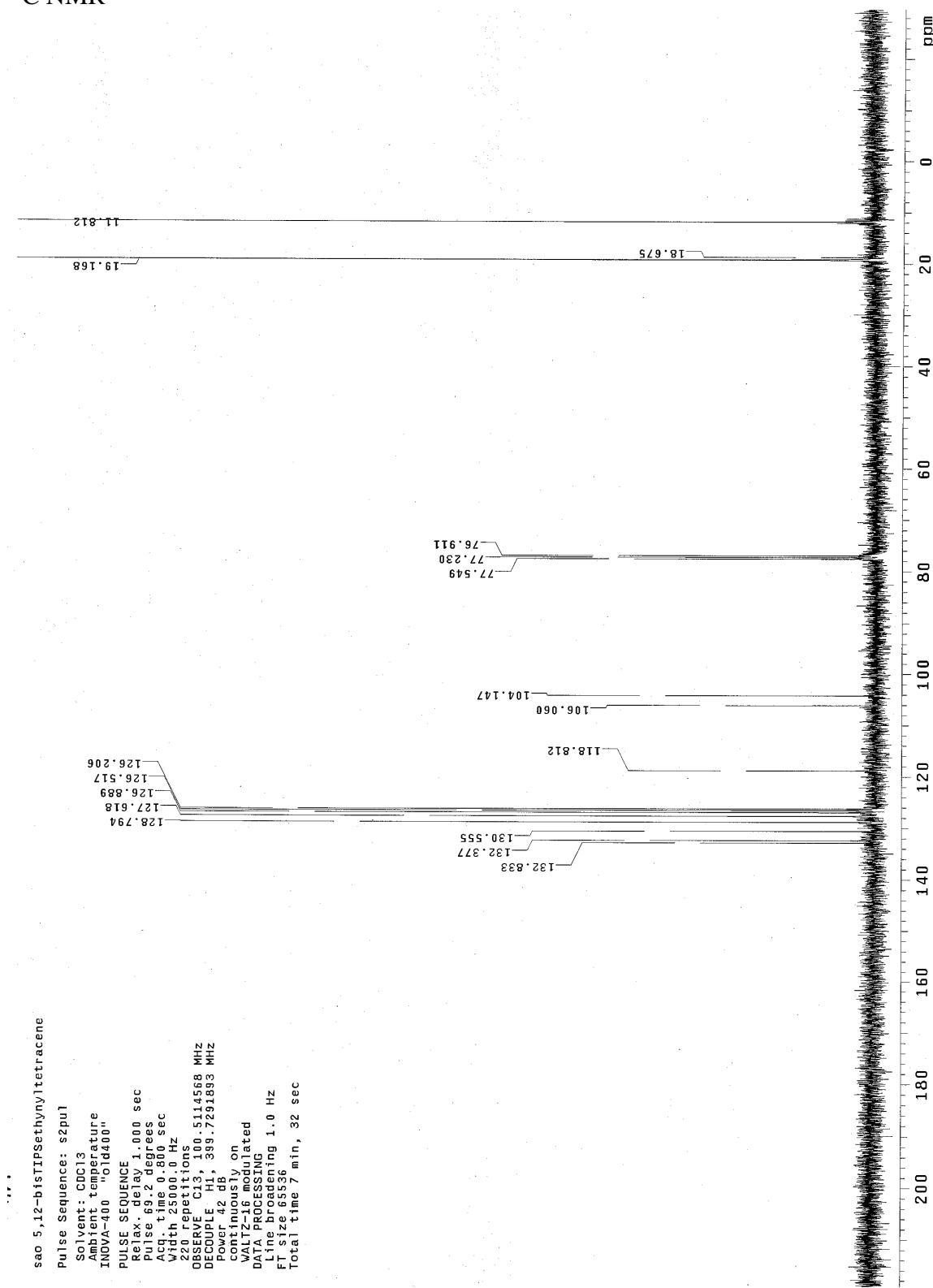
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pulse Sequence: sapl
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  Ambient temperature
  INNOVA-300 "Old400"
  PULSE SEQUENCE
  Relax delay 1.000 sec
  pulse 71.2 degrees
  Acq. time 3.744 sec
  width 6000.6 Hz
  8 rep/iterations
  OBSERVE H1, 399.7721745 MHz
  DATA PROCESSING
  size 65536
  Total time 2 min, 32 sec

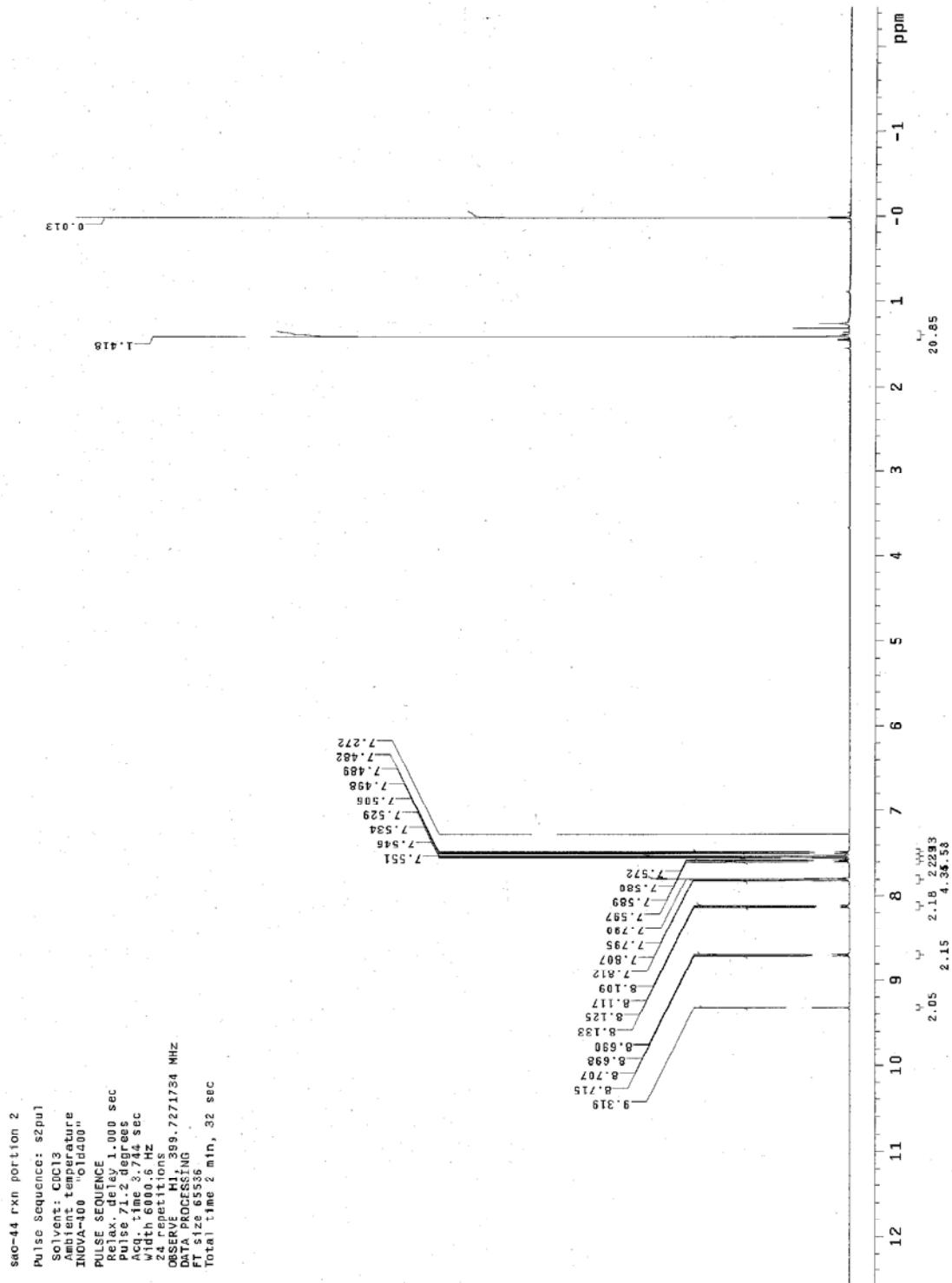
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## Compound 3 $^{13}\text{C}$ NMR



## Compound 4 <sup>1</sup>H NMR



Compound 4  
<sup>13</sup>C NMR

sao bis-5,12-t-butylphenyl ethynyltetraacene

Pulse Sequence: s2pul

Solvent: CDCl<sub>3</sub>  
 Ambient temperature  
 INGLA-400 "Orid400"

PULSE SEQUENCE

Relax. delay 1.000 sec

Pulse 69.2 degrees

Acq. time 0.800 sec

Width 2500.0 Hz

250 repetitions

OBSERVE C13, 100.5114568 MHz

DECOUPLE H1, 100.5114568 MHz

Power 42 dB

continuously on

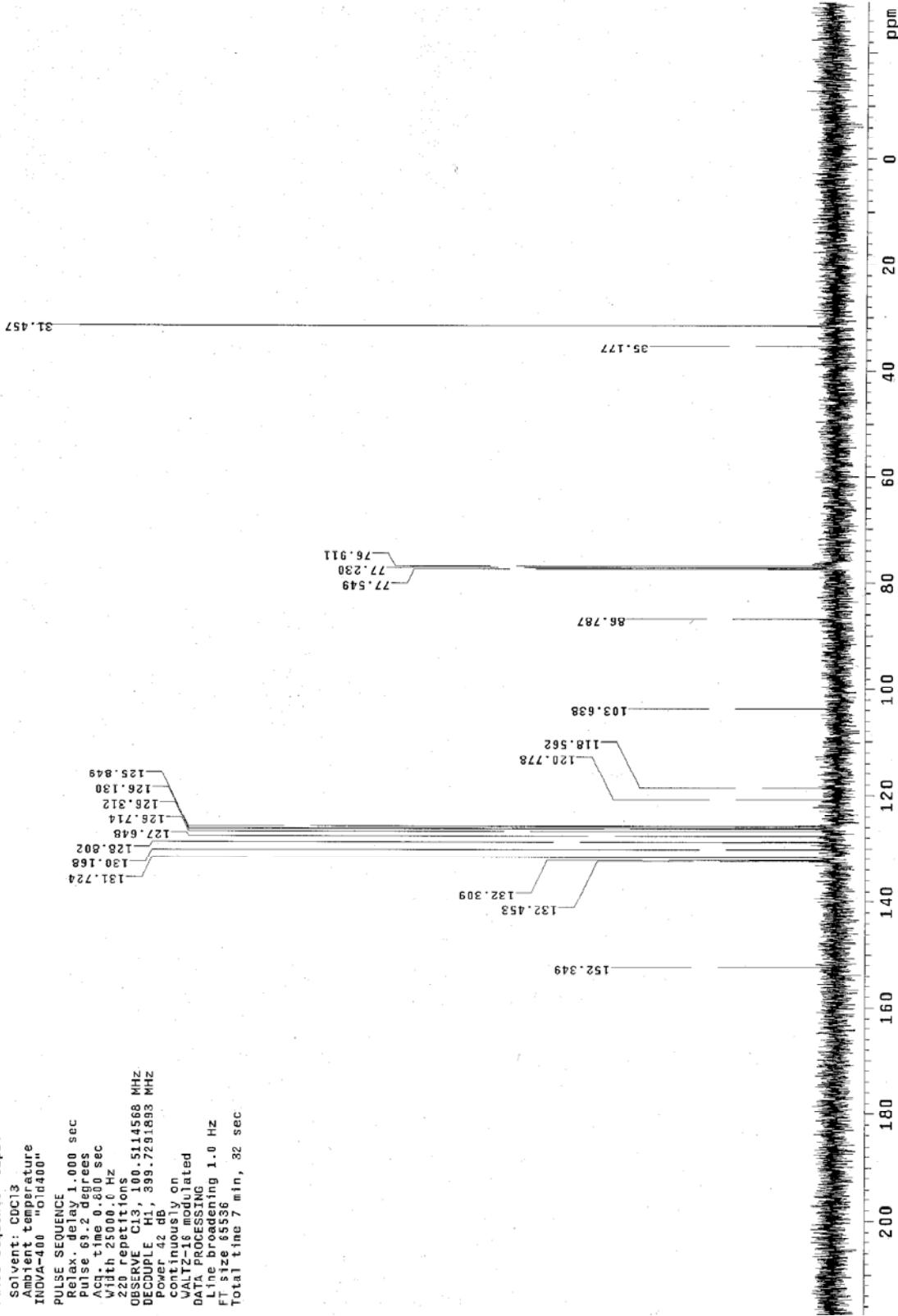
WALTZ16 modulated

DATA PROCESSING

Line broadening 1.0 Hz

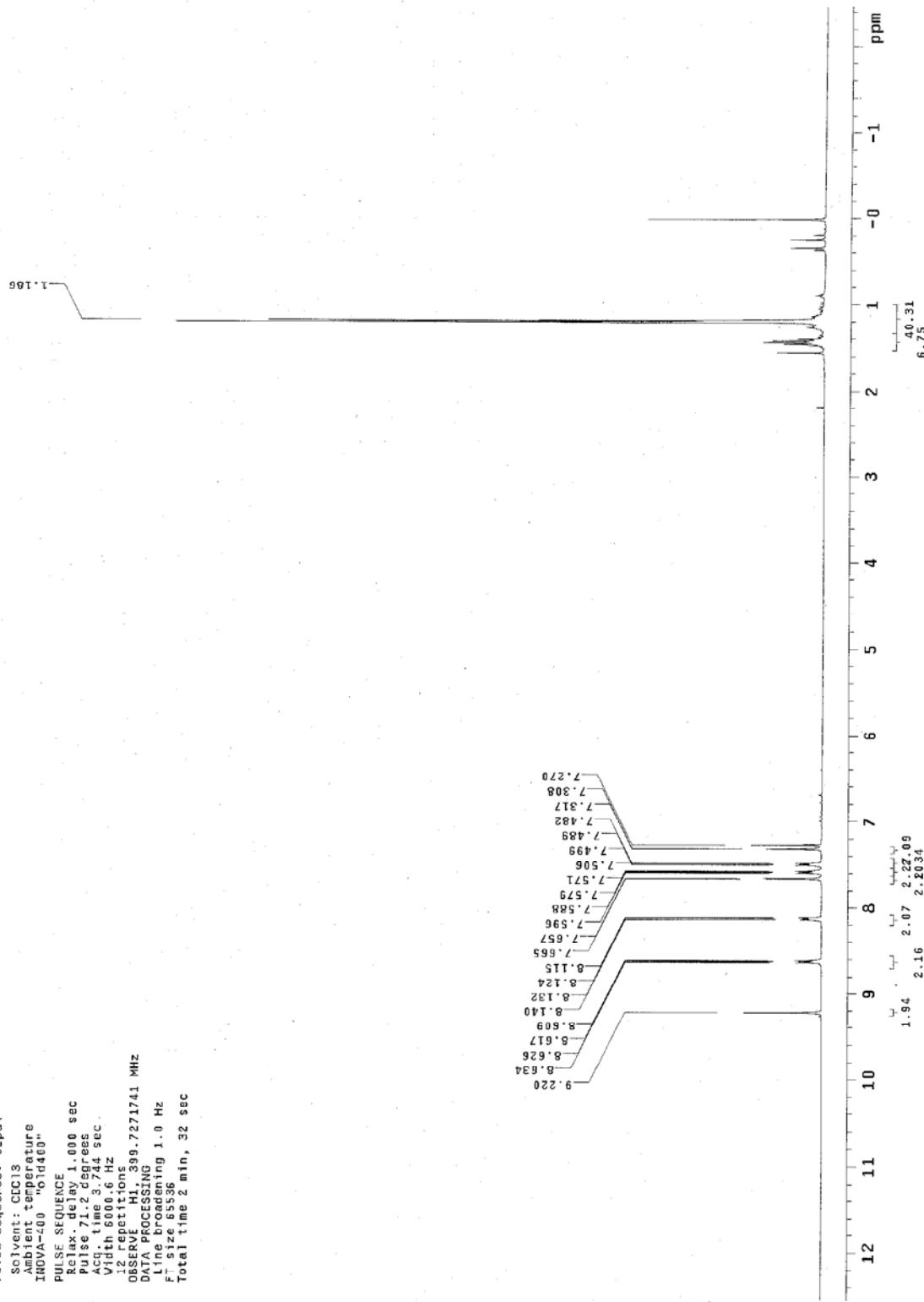
FT size 65536

Total time 7 min, 32 sec

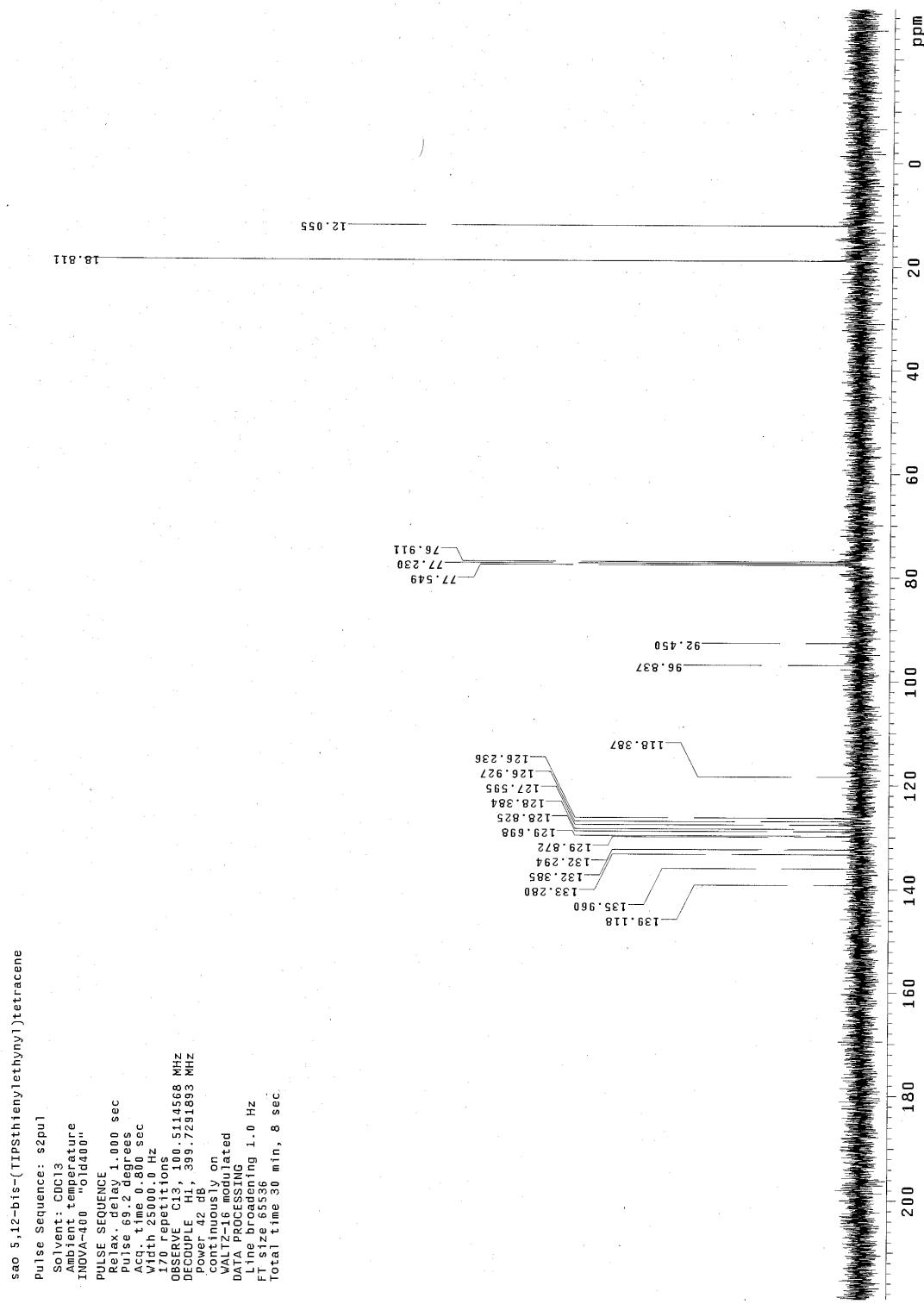


Compound 5  
<sup>1</sup>H NMR

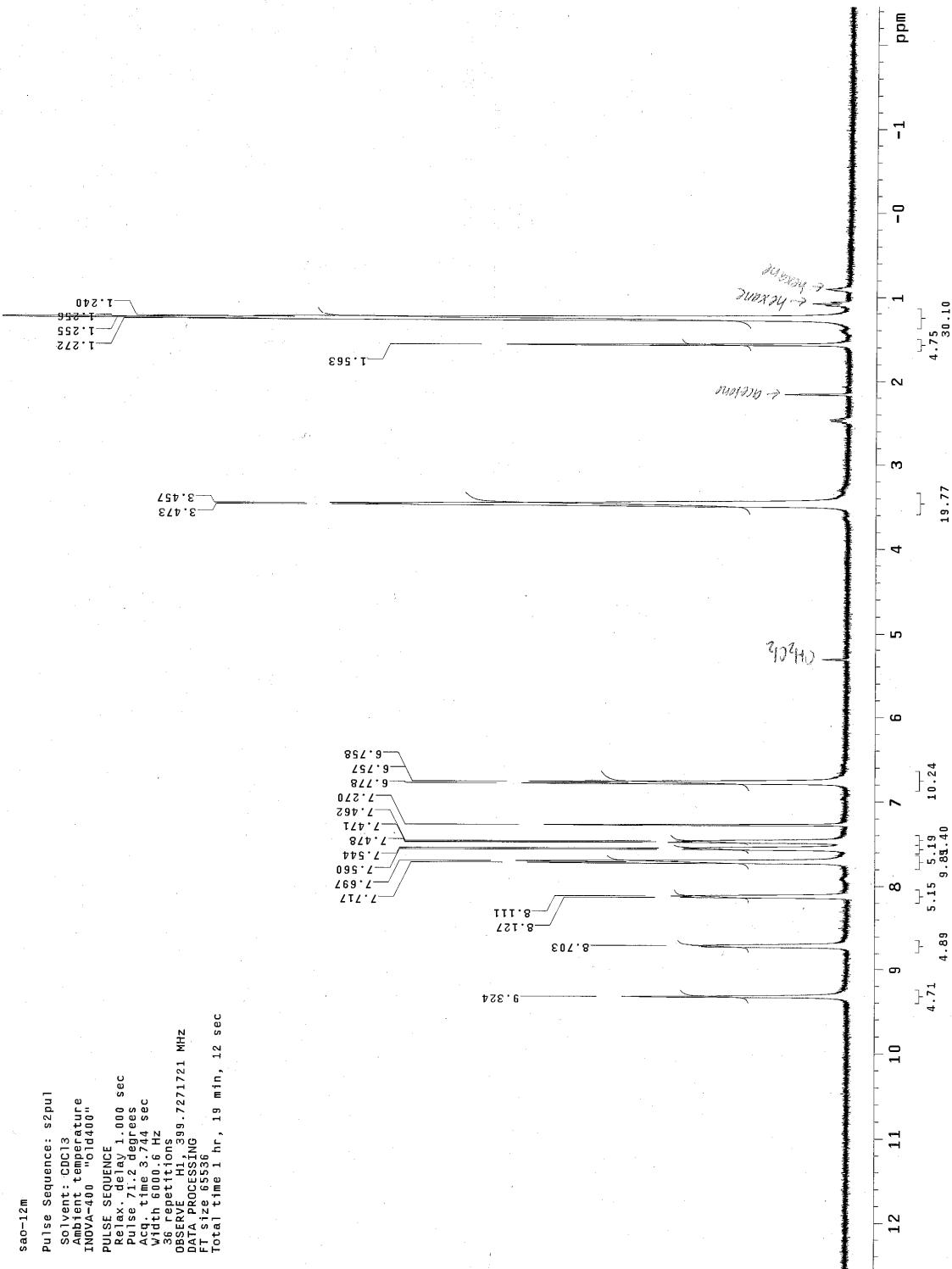
sao 5,12-bis-(TIPS thiényl éthynyl) tetracène  
Pulse Sequence: s2pu  
Solvent: CCl<sub>4</sub>  
Ambient temperature  
TROVA=0.0 "Old400"  
PULSE SEQUENCE  
Relax. delay 1.000 sec  
Pulse 71.2 degrees  
Acc. time 3.244 sec  
Width 6000.0 Hz  
12.8 sec per 1000 points  
OBSERVE H<sub>1</sub>, 399.7271741 MHz  
DATA PROCESSING  
Line broadening 1.0 Hz  
F size 6536  
Total time 2 min, 32 ssc



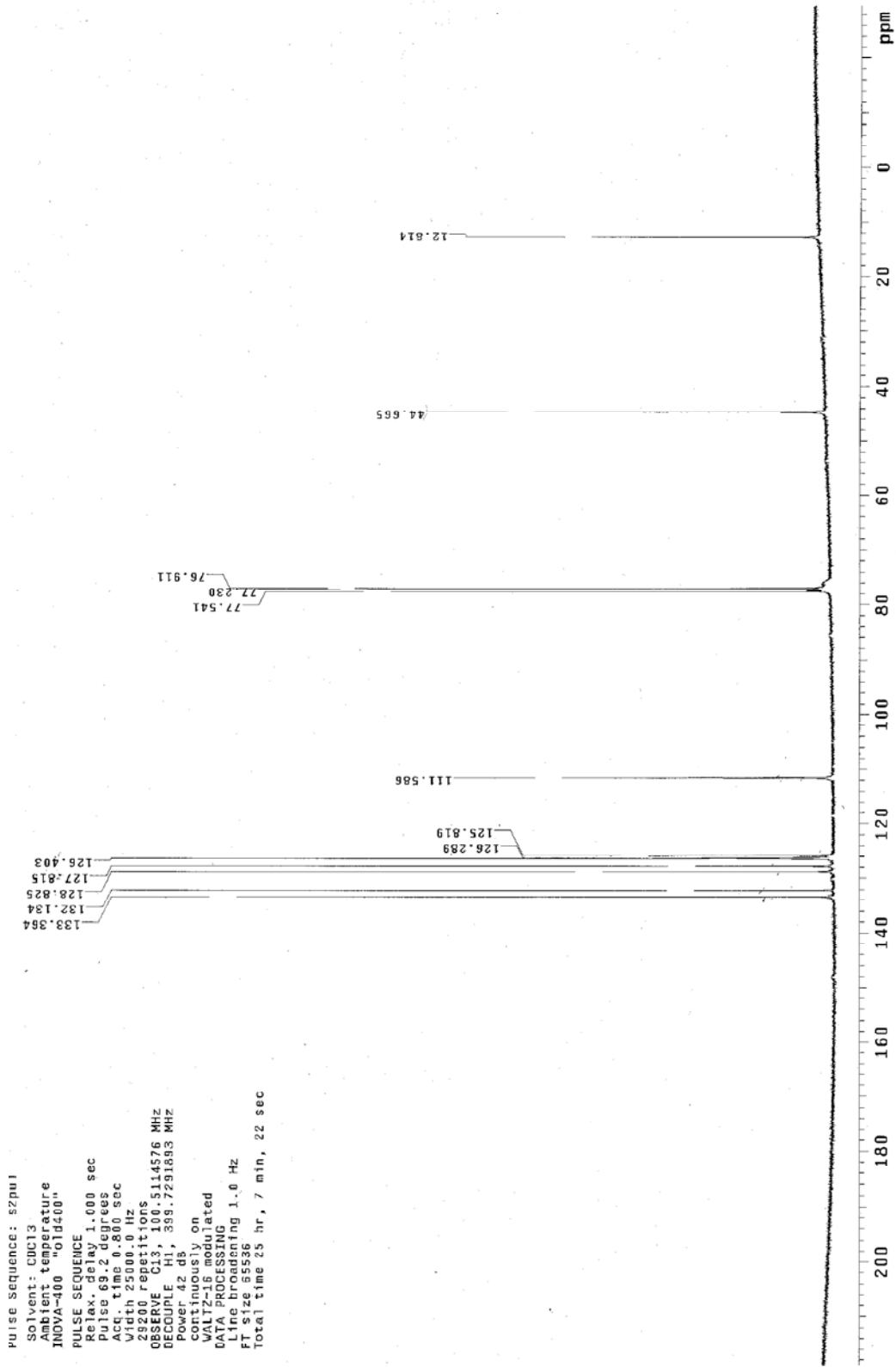
## Compound 5 $^{13}\text{C}$ NMR



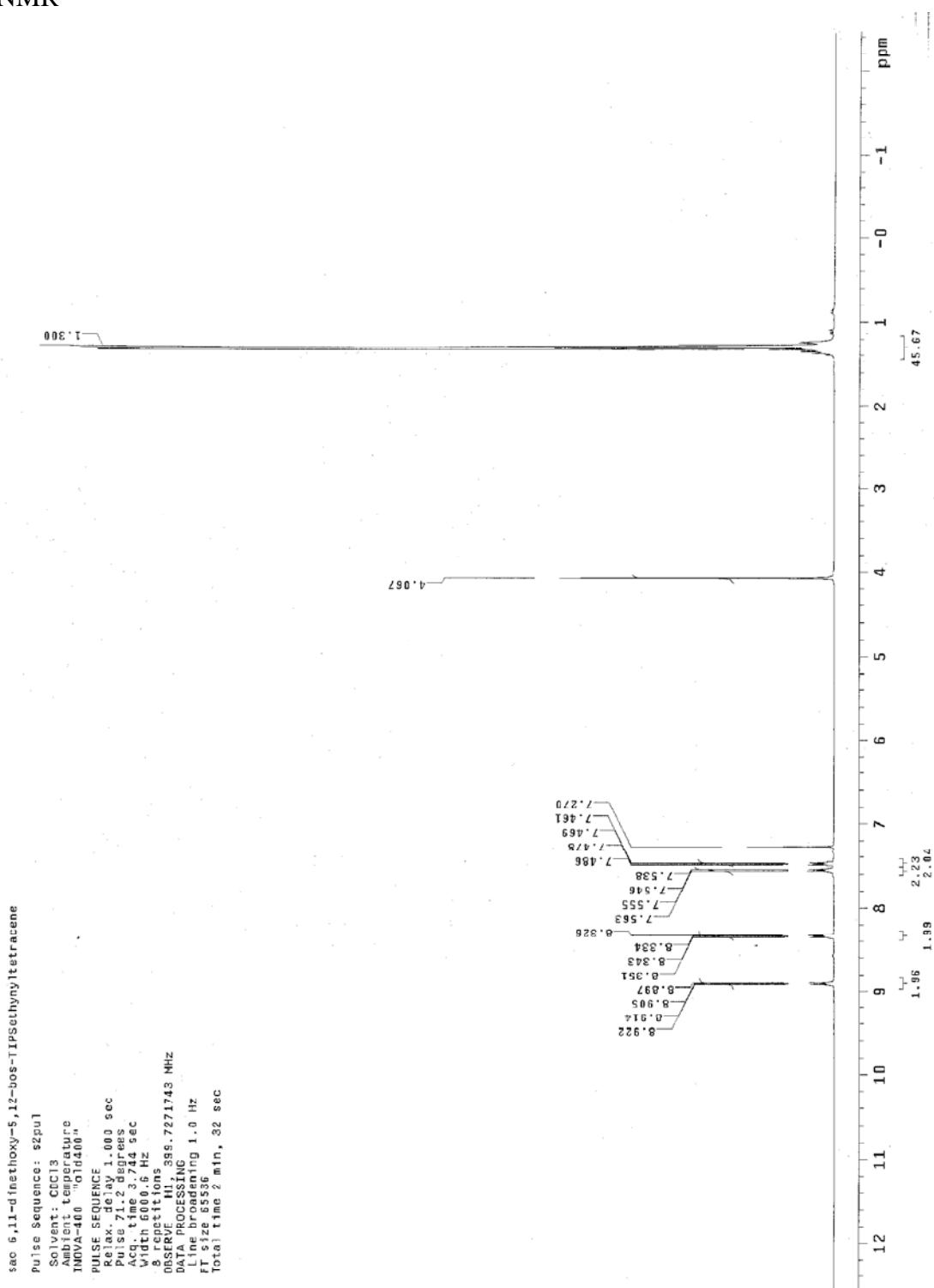
Compound 6  
<sup>1</sup>H NMR



Compound 6  
<sup>13</sup>C NMR

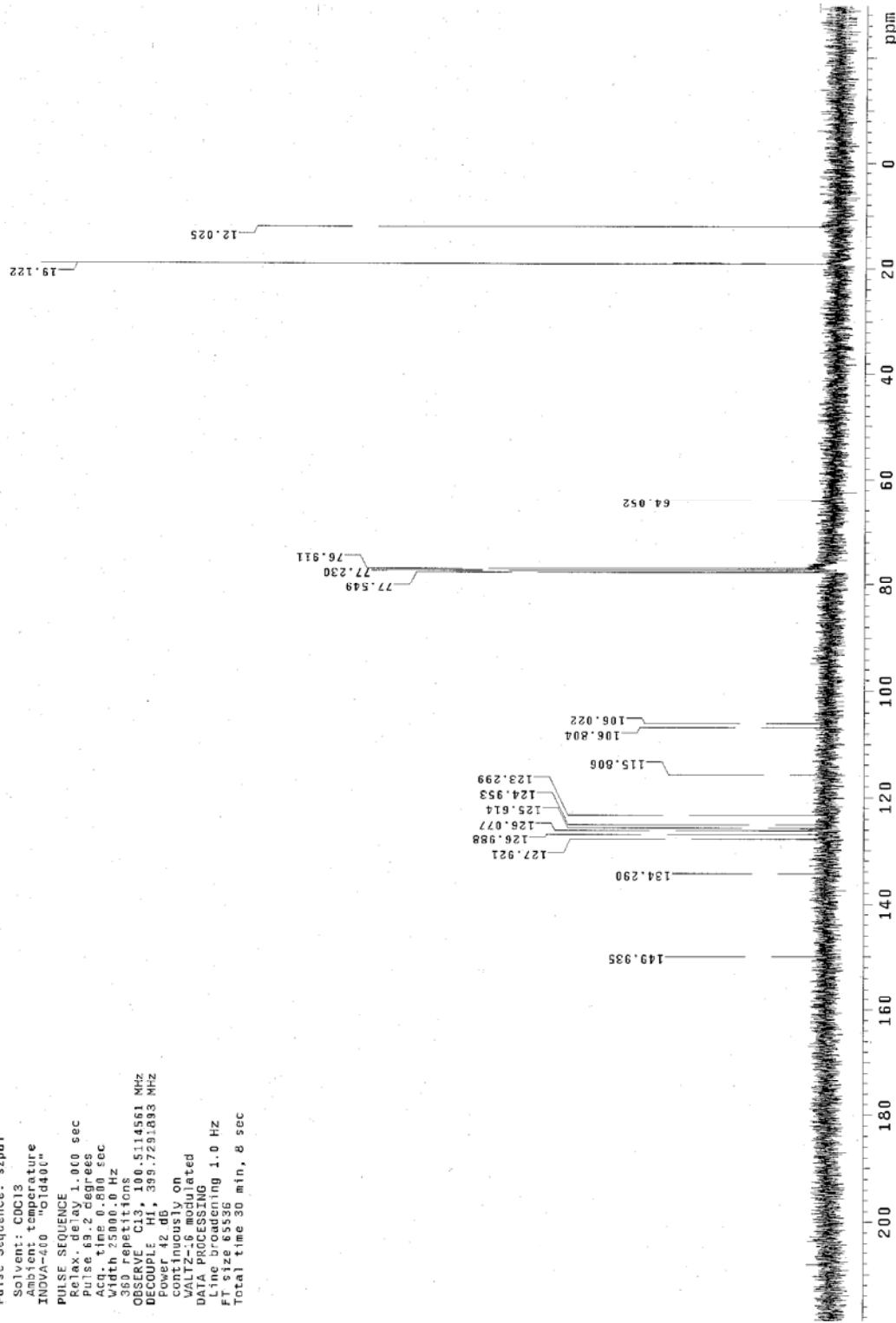


Compound 7  
<sup>1</sup>H NMR

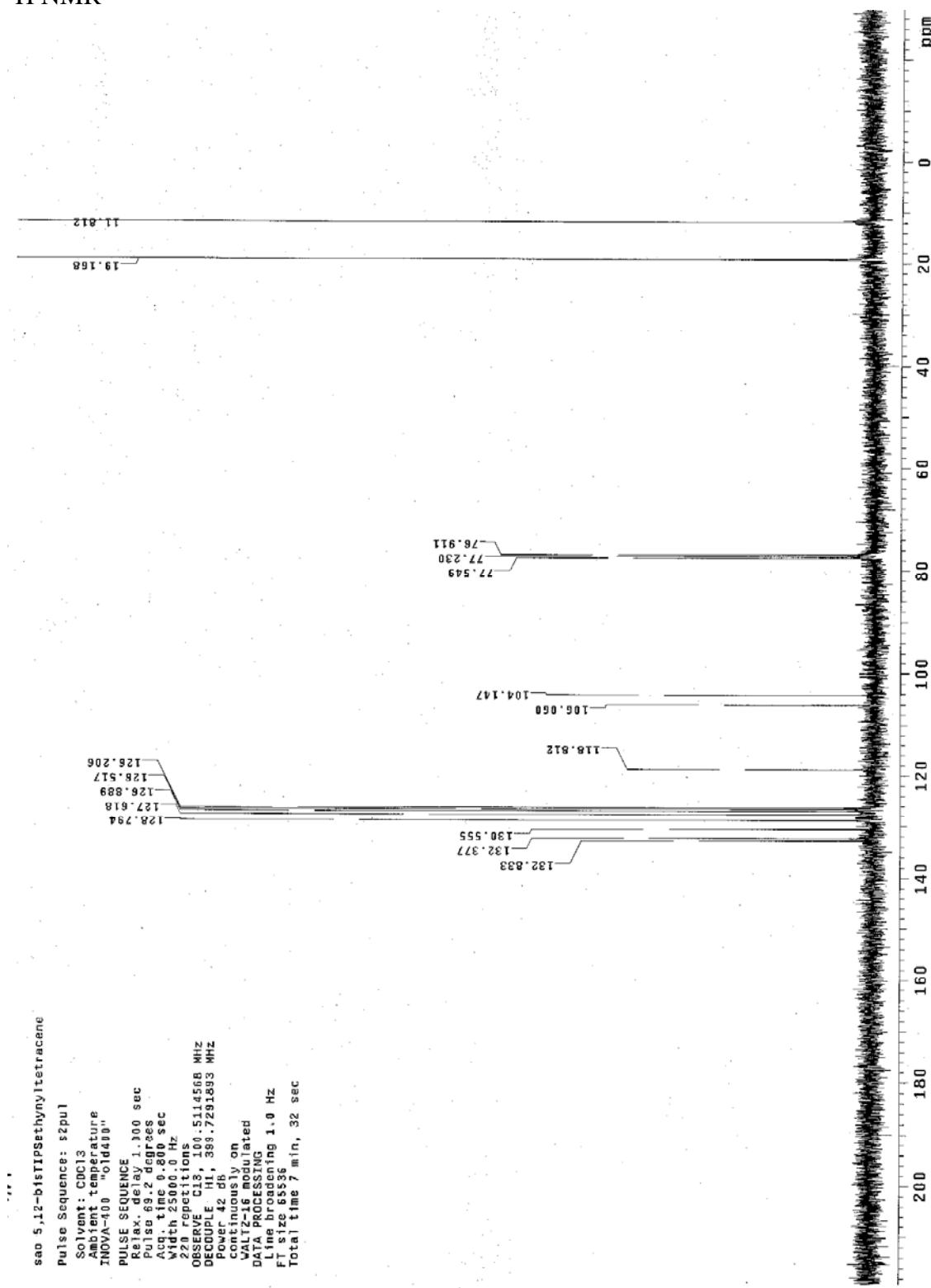


Compound 7  
<sup>13</sup>C NMR

sao dmethoxy-bis[1H]Sethynyl]tetqu[inone  
 Pulse Sequence: s2pul  
 Solvent: CDCl<sub>3</sub>  
 Ambient temperature  
 INDOA-400 "Poldaq"  
 PULSE SEQUENCE  
 Relax. delay 1.00 sec  
 Pulse 69.2 degrees  
 Acq. time 0.800 sec  
 Width 2500.0 Hz  
 360 repetitions  
 OBSERVE C13, 100.5114561 MHz  
 DECOUPLE H1, 399.7201283 MHz  
 Power 42 dB  
 WALTZ-16 modulated  
 DATA PROCESSING  
 Line broadening 1.0 Hz  
 FT size 65536  
 Total time 30 min, 8 sec



Compound 8  
<sup>1</sup>H NMR



## Compound 8 <sup>13</sup>C NMR

